

Reaction Rates and Mechanisms Quiz

Name: _____

____K

Assessment of Learning (Product)

20

Multiple Choice; Circle the most correct answer. (20K)

1. The following property can be measured to determine the rate of the reaction.

- A) change in mass B) change in colour C) change in volume
D) change in pressure E) all of the above depending on the reaction

2. Given the following reaction: $\text{H}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightarrow 2 \text{HCl}$ As this reaction takes place:

- A) H_2 and Cl_2 are consumed at the same rate
B) HCl is produced at the same rate that H_2 is consumed
C) HCl is produced at the half the rate that Cl_2 is consumed
D) HCl is produced at the same rate that Cl_2 is consumed

3. The slowest of the following reactions is:

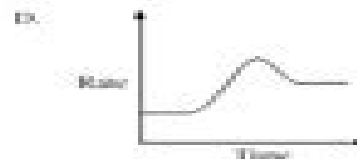
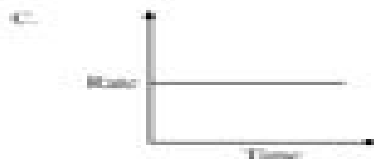
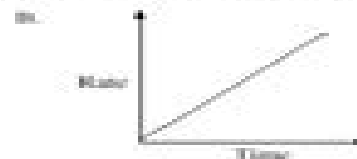
- A) $\text{Ag}_{(\text{aq})}^+ + \text{Cl}_{(\text{aq})}^- \rightarrow \text{AgCl}_{(\text{s})}$
B) $\text{H}_2\text{O}_{(\text{aq})}^+ + \text{OH}_{(\text{aq})}^- \rightarrow 2\text{H}_2\text{O}$
C) $3\text{Ba}^{2+}_{(\text{aq})} + 2\text{PO}_4^{3-}_{(\text{aq})} \rightarrow \text{Ba}_3(\text{PO}_4)_2_{(\text{s})}$
D) $\text{Cu}_{(\text{s})} + 2\text{Ag}^{+}_{(\text{aq})} \rightarrow \text{Cu}^{2+}_{(\text{aq})} + 2\text{Ag}_{(\text{s})}$

4. The average kinetic energy of colliding particles can be increased by;

- A) adding a catalyst. B) increasing pressure
C) increasing temperature D) increasing reactant concentration.

5. Consider the following reaction: $2\text{H}_2\text{O}_{2(\text{g})} \rightarrow 2\text{H}_2\text{O}_{(\text{g})} + \text{O}_{2(\text{g})}$

Which graph shows the relationship between rate of consumption of H_2O_2 and time?



Reaction Rates Assessment Answers

Joanna Hayden



Reaction Rates Assessment Answers:

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