

THIRD TERM EXAMINATION QUESTION

CHEMISTRY: SSS II

1. Consider the following reaction equation: $C_2H_{4g} + 3O_{2g} \rightarrow 2CO_{2g} + 2H_2O_{lg}$. The volume of oxygen at s.t.p. that will be required to burn 14g of ethane is? [$C_2H_4 = 28$, Molar volume of gas at s.t.p. = 22.4 dm^3] (a) 67.2 dm^3 (b) 33.6 dm^3 (c) 11.2 dm^3 (d) 3.73 dm^3
2. 50cm^3 of sulphur (IV) oxide, 800cm^3 of ammonia, 450cm^3 of hydrogen chloride, 1.0cm^3 of carbon (IV) oxide will respectively saturate 1.0cm^3 of water at 15°C . Which of the following is suitable for demonstrating the fountain experiment? (a) Sulphur (IV) oxide and hydrogen chloride (b) carbon (IV) oxide and ammonia (c) Ammonia and hydrogen chloride (d) carbon (IV) oxide and sulphur (IV) oxide
3. The volume of 0.25 mol dm^{-3} solution of KOH that would yield 6.5g of solid KOH on evaporation is? [$K = 39$, $O = 16$, $H = 1$] (a) 464.30 cm^3 (b) 625 cm^3 (c) 1000 cm^3 (d) 2153.80 cm^3
4. Solubility is practically applied in (a) fractional distillation (b) solvent extraction (c) the determination of pH (d) the determination of saturation in hydrocarbons
5. The following factors affect the solubility of a solid in a given solvent **EXCEPT** (a) nature of solute (b) nature of solvent (c) pressure (d) temperature



- Which of the following deductions could be **correctly** made from the graph? (a) the solubility of U is not affected by change in temperature (b) the solubility of S decreases with increasing temperature (c) T is most soluble among the salts (d) S is least soluble among the salts
7. Soap lather in is an example of a colloid in which a (a) liquid is dispersed in gas (b) solid is dispersed in liquid (c) gas is dispersed in liquid (d) liquid is dispersed in liquid
 8. The number of sulphur atoms in 3.20 g of SO_{2g} is? [$O = 16$, $S = 32$, Avogadro's constant = 6.02×10^{23}] (a) 3.01×10^{23} (b) 6.02×10^{23} (c) 6.02×10^{23} (d) 1.20×10^{24}
 9. At 27°C , 58.5g of sodium chloride is present in 250cm^3 of a solution. The solubility of sodium chloride at this temperature is? [Molar mass of sodium chloride = 111.0g mol^{-1}] (a) 2.11 mol dm^{-3} (b) 0.25 mol dm^{-3} (c) 1.00 mol dm^{-3} (d) 0.50 mol dm^{-3}
 10. Which of the following pollutants will lead to the depletion of ozone layer? (a) chlorofluorocarbon (b) carbon (II) oxide (c) hydrogen sulphide (d) sulphur (IV) oxide
 11. Consider the equation: $Cr_2O_7^{2-} + 6Fe^{2+} + 14H^+ \rightarrow 2Cr^{3+} + 6Fe^{3+} + 7H_2O$. The oxidation number of chromium changes from (a) $+5$ to $+3$ (b) $+6$ to $+3$ (c) -2 to $+3$ (d) $+7$ to $+3$

Ss1 Test Question On Chemistry

Jicheng Xie



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