

**COMMUNITY SECONDARY SCHOOL, UMULUMGBE**

**UDI LGA, ENUGU STATE.**

**THIRD TERM EXAMINATION 2023/2024 ACADEMIC SESSION**

**SUBJECT: CHEMISTRY CLASS: SS 1 TIME: 1½ HOURS DATE: 03/07/2024**

**Instruction: Answer all questions in section A and three in B**

**SECTION A (Objective Test Questions: Choose correct option from A-D)**

- Which of the following compounds is a basic salt? (A)  $Mg(NO_3)_2$  (B)  $(NH_4)_2SO_4$  (C)  $K_4Fe(CN)_6$  (D)  $Zn(OH)Cl$
- How many protons does  $^{40}_{20}Ca$  contains? (A) 20 (B) 30 (C) 40 (D) 60
- Which of the following expressions correctly represents Charles' Law? (A)  $PV = K$  (B)  $R \propto \frac{1}{V}$  (C)  $PV = RT$  (D)  $V = kT$
- What is the value of  $-14^\circ C$  on the Kelvin temperature scale? (A) 259K (B) 259K (C) 287 (D) 287°K
- Which of the following gases has the lowest rate of diffusion under the same condition? [H=1; He=4; O=16; Cl=35.5] (A)  $Cl_2$  (B)  $H_2$  (C)  $He$  (D)  $O_2$
- Which of the following substances with the corresponding pH value is an alkaline?

	Substance	pH
A	Lime	2.4
B	Banana	4.6
C	Distilled water	7.0
D	Milk of magnesia	10.5

- Which of the following salts on hydrolysis would give an alkaline solution? (A)  $NH_4Cl$  (s) (B)  $AlCl_3$  (s) (C)  $CH_3COONa$  (s) (D)  $NH_4NO_3$  (s)
- Which of the following processes do NOT take place in domestic water treatment? (A) Chlorination (B) Flocculation (C) Neutralization (D) Sedimentation
- Consider the following reaction equation  $H_3O^+ + OH^- \longrightarrow 2H_2O(l)$   
The reaction represents (A) esterification (B) hydrolysis (C) neutralization (D) redox reaction
- Which of the following statements is NOT a chemical property of an acid? (A) Evolution of ammonia gas when heated with ammonium salts (B) Evolution of  $CO_2$  when added to a trioxocarbonate (IV) salt (C) Formation of salt and water with alkalis (D) Formation of salt and hydrogen gas with reactive metals.
- Tetraoxosulphate (VI) acid is described as strong acid because it is highly (A) reactive (B) concentrated (C) corrosive (D) ionized in water
- During water treatment for town supply, water is passed through layers of sand beds in order to (A) soften the water (B) filter the water (C) decolourize the water (D) destroy the germs in the water
- The basicity of tetraoxosulphate (VI) acid is (A) 1 (B) 2 (C) 3 (D) 4
- The mathematical expression of pH is (A)  $\text{Log}_{10}\left[\frac{1}{H_3O^+}\right]$  (B)  $\text{Log}_{10}(H_3O^+)$  (C)  $\text{Log}_{10}\left[\frac{1}{OH^-}\right]$  (D)  $\text{Log}_{10}(OH^-)$
- Two atoms represented as  $^{234}_{92}U$  and  $^{238}_{92}U$  are (A) Isomers (B) allotropes (C) Isotopes (D) anomers
- The number electrons in the valence shell of an element of atomic number 14 is (A) 1 (B) 2 (C) 3 (D) 4

# Ss1 Chemistry Practical 2015 3rd Term

**Huangqi Zhang**



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