

Chemical Formula Writing Worksheet Two

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

| Cations | | | | | | |
|----------|-------------------|-----------|------------|---------|--------|-----------|
| Anions | zinc | iron (II) | iron (III) | gallium | silver | lead (IV) |
| chloride | ZnCl ₂ | | | | | |
| acetate | | | | | | |
| nitrate | | | | | | |
| oxide | | | | | | |
| nitride | | | | | | |
| sulfate | | | | | | |

Write the formulas for the following compounds:

- 1) copper (II) chloride _____
- 2) lithium acetate _____
- 3) vanadium (III) selenide _____
- 4) manganese (IV) nitride _____
- 5) beryllium oxide _____
- 6) sodium sulfate _____
- 7) aluminum arsenide _____
- 8) potassium permanganate _____
- 9) chromium (VI) cyanide _____
- 10) tin (II) sulfite _____
- 11) vanadium (V) fluoride _____
- 12) ammonium nitrate _____

Staff Insa Edu Chemical Formula

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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