

SS1 CHEMISTRY EXAM QUESTIONS FOR SECOND TERM
PAPER 1: TEST OF OBJECTIVE QUESTIONS

1. If the valency of hydrogen is one and that of oxygen is two the formula is
(a) HO_2 (b) H_2O (c) 2OH (d) H_2O_2
2. The IUPAC name for the compound $\text{ClCH}_2\text{CH}_2\text{OH}$ is
(a) 1-chloropropanol (b) 1-chloropropanol (c) 1-chloroethanol (d) 2-chloroethanol
3. Petroleum is a non-renewable source of energy because it
(a) is formed naturally (b) is cheap (c) can be recycled after use (d) cannot be regenerated once used up
4. One of the Isotopes of hydrogen is (a) Ozone (b) Deuterium (c) diamond
(d) graphite
5. Which of the following arrangements shows increasing order of reactivity of the halogens?
(a) $\text{F}_2 > \text{Cl}_2 > \text{Br}_2 > \text{I}_2$ (b) $\text{I}_2 < \text{Br}_2 < \text{Br}_2 < \text{F}_2$ (c) $\text{F}_2 < \text{Cl}_2 < \text{Br}_2 < \text{I}_2$ (d) $\text{I}_2 > \text{Br}_2 > \text{Cl}_2 > \text{F}_2$
6. Hydrogen is prepared in the laboratory by the action of dilute H_2SO_4 on the following metals except (a) Zn (b) Na (c) Al (d) Fe
7. The gas given out when sodium reacts with water is (a) oxygen (b) hydrogen (c) nitrogen
(d) Chlorine
8. The minimum amount of energy required for effective collisions between reacting particles is known as (a) activation energy (b) bond energy (c) kinetic energy (d) potential energy
9. On a large scale, hydrogen is produced from (a) producer gas (b) water gas (c) coke (d) coal tar
10. In which group is oxygen in the periodic table? (a) IV (b) VI (c) VII (d) IV
11. The bond formed between H_2O and H^+ to form hydroxonium ion H_3O^+ is
(a) dative (b) covalent (c) electrovalent (d) sodium
12. Which of the following is an allotrope of oxygen? (a) oxide (b) Ozone
(c) diamond (d) graphite
13. Oxygen is prepared by heating. (a) Na_2CO_3 (b) KClO_3 (c) K_2SO_4 (d) H_2S

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