

TOP CROWN PROGRESSIVE PATH ACADEMY  
39/40 GANIU ODUBANWO STREET POTA BADAGRY

THIRD TERM EXAMINATION

SUBJECT: CHEMISTRY

CLASS: S. S. S ONE

DURATION: 2HR

1. Acid turns blue litmus paper \_\_\_\_\_?  
a. yellow    b. colourless    c. green    d. red
2. Acids are characterized by    a. Bitter taste    b. soapy taste  
c. sour taste    d. all listed
3. An element existing in two or more physical states and maintaining the same chemical characteristics is called \_\_\_\_\_ a. polymerization    b. allotropes  
c. isotopes    d. crystal
4. Which allotrope of carbon is used in lead pencil?  
a. graphite    b. coke    c. coal    d. diamond
5. Carbon dioxide is a gas that burns with \_\_\_\_\_ flame  
a. yellow    b. green    c. blue    d. colourless
6. Hydrocarbons are of two classes namely: a. graphite and diamond  
b. coke and coal    c. aliphatic and gas    d. aromatic and aliphatic
7. An example of aromatic hydrocarbons is    a. Ethane    b. methane  
c. cyclopropane    d. benzene
8. Which is not a raw material for chemical industry    a. petroleum  
b. coal    c. alum    d. salt
9. Which is not a use of hydrogen trioxocarbonate    a. As baking powder  
b. As health salts    b. As fire extinguishers    d. To manufacture hydroxide
10. Which is not a property of carbon(II)oxide    a. It is colourless  
b. It is slightly lighter than air    c. It is insoluble in water    d. It is soluble
11. Which is a type of amorphous    a. carbon black charcoal    b. plant charcoal  
c. red charcoal    d. graded charcoal
12. Which of these is not a type of coal?    a. peat    b. lignite bituminous    c. coal
13. Coke is a non-volatile residue, this statement is    a. true    b. false    c. negative
14. Water gas is a mixture that contains CO and \_\_\_\_\_ gas?    a. hydrogen  
b. helium    c. lithium    d. berlium
15. The number of replaceable hydrogen atom in one molecule of an acid indicates its  
a. Basicity    b. Alkalinity    c. Acidity    d. Reactivity
16. Which of the following is /are characteristics properties of acids? They i. form hydroxonium ions in water ii. liberate carbon (iv)oxide from trioxocarbonate(iv) salts. iii. Turn blue litmus paper red. iv. have a soapy feel  
a. i, iv only    b. i and ii    c. iii and iv    d. i, ii, and iii
17. The following acid are monobasic except    a. Trioxonitrate(v)acid    b. Hydrochloric acid  
c. Ethanoic acid    d. Tetraoxosulphate (v)acid
18. Which of the following is an acid salt?    a.  $(\text{NH}_4)_2\text{CO}_3$     b.  $\text{CH}_3\text{COONa}$     c.  $\text{KHSO}_4$     d.  $\text{MgSO}_4$
19. Alums are classified as    a. basic salts    b. acid salts    c. anhydrous salts    d. double salts
20. Which of the following conclusion about solution of pH4 is correct  
a. it contains more  $\text{OH}^-$  than  $\text{H}_3\text{O}^+$     b. its POH value will be 10    c. it is more acidic than a solution of pH2

# Ss Chemistry Exam

**Mansoor Niaz**



## **Ss Chemistry Exam:**

Chemistry Education and Contributions from History and Philosophy of Science Mansoor Niaz, 2015-12-23 This book explores the relationship between the content of chemistry education and the history and philosophy of science HPS framework that underlies such education It discusses the need to present an image that reflects how chemistry developed and progresses It proposes that chemistry should be taught the way it is practiced by chemists as a human enterprise at the interface of scientific practice and HPS Finally it sets out to convince teachers to go beyond the traditional classroom practice and explore new teaching strategies The importance of HPS has been recognized for the science curriculum since the middle of the 20th century The need for teaching chemistry within a historical context is not difficult to understand as HPS is not far below the surface in any science classroom A review of the literature shows that the traditional chemistry classroom curricula and textbooks while dealing with concepts such as law theory model explanation hypothesis observation evidence and idealization generally ignore elements of the history and philosophy of science This book proposes that the conceptual understanding of chemistry requires knowledge and understanding of the history and philosophy of science Professor Niaz's book is most welcome coming at a time when there is an urgently felt need to upgrade the teaching of science The book is a huge aid for adding to the usual way presenting science as a series of mere facts also the necessary mandate to show how science is done and how science through its history and philosophy is part of the cultural development of humanity Gerald Holton Mallinckrodt Professor of Physics these are valence bond VB and molecular orbital MO theories He emphasizes the importance of having rival theories and interpretations in science and its advancement He further argues that this VB MO rivalry is still alive and together the two conceptual frames serve as the tool kit for thinking and doing chemistry in creative manners The author surveys chemistry textbooks in the light of the how the books preserve or not the balance between the two theories in describing various chemical phenomena This Talmudic approach of conceptual tension is a universal characteristic of any branch of evolving wisdom As such Mansoor's book would be of great utility for chemistry teachers to examine how can they become more effective teachers by recognizing the importance of conceptual tension Sason Shaik Saeree K and Louis P Fiedler Chair in Chemistry Director The Lise Meitner Minerva Center for Computational Quantum Chemistry The Hebrew University of Jerusalem ISRAEL

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potential by informing them about how thought processes work while also offering practical strategies and techniques This book not only offers a deep understanding of the learning process but also offers readers practical steps to improve their cognitive abilities If you want to discover and develop your learning potential this book will be an indispensable guide for you

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**University of Michigan Official Publication** ,1951      **Peterson's Grad Programs in Physical Sciences, Math, Ag Sciences, Envir & Natural Res 20154 (Grad 4)** Peterson's,2014-10-21 Graduate Programs in the Physical Sciences Mathematics Agricultural Sciences the Environment Natural Resources 2015 contains more than 3 000 graduate programs in the relevant disciplines including agriculture and food sciences astronomy and astrophysics chemistry physics mathematics environmental sciences and management natural resources marine sciences and more Informative data profiles for more than 3 000 graduate programs at nearly 600 institutions are included complete with facts and figures on accreditation degree requirements application deadlines and contact information financial support faculty and student body profiles Two page in depth descriptions written by featured institutions offer complete details on specific graduate programs schools or departments as well as information on faculty research Comprehensive directories list programs in this volume as well as others in the graduate series      One Hundred Years at the Intersection of Chemistry and Physics Jeremiah James,Thomas Steinhauser,Dieter Hoffmann,Bretislav Friedrich,2011-10-27 This volume occasioned by the centenary of the Fritz Haber Institute formerly the Institute for Physical Chemistry and Electrochemistry covers the institute s scientific and institutional history from its founding until the present The institute was among the earliest established by the Kaiser Wilhelm Society and its inauguration was one of the first steps in the development of Berlin Dahlem into a center for scientific research Its establishment was made possible by an endowment from Leopold Koppel granted on the condition that Fritz Haber well known for his discovery of a method to synthesize ammonia from its elements be made its director The history of the institute has largely paralleled that of 20th century Germany It undertook controversial weapons research during World War I followed by a Golden Era during the 1920s in spite of financial hardships Under the National Socialists it experienced a purge of its scientific staff and a diversion of its research into the service of the new regime accompanied by a breakdown in its international relations In the immediate aftermath of World War II it suffered crippling material losses from which it

recovered slowly in the post war era In 1953 shortly after taking the name of its founding director the institute joined the fledgling Max Planck Society During the 1950s and 60s the institute supported diverse researches into the structure of matter and electron microscopy in a territorially insular and politically precarious West Berlin In subsequent decades as both Berlin and the Max Planck Society underwent significant changes the institute reorganized around a board of coequal scientific directors and a renewed focus on the investigation of elementary processes on surfaces and interfaces topics of research that had been central to the work of Fritz Haber and the first Golden Era of the institute

*Super 10 Practice Sets for SSC CGL Mains (Tier II) Exam with 2 Online Sets* Disha Experts,2019-07-02 Super 10 Practice Sets for SSC CGL Mains Tier II Exam with 2 Online Sets is extensively prepared for the students who are preparing for the Combined Graduate Level CGL II Examination Each Set contains 2 sections Quantitative Ability 100 Questions English Language and Comprehension 200 Questions This book will help the students in understanding the structure and format of the exam The book provides a total of 10 Practice Sets 8 in Book and 2 Online which are created according to the latest format and guidelines set by Staff Selection Commission The Solutions are provided to each and every question in the book

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this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

*Culture of Chemistry* Balazs Hargittai,István Hargittai,2015-04-20 Includes specially selected articles that previously appeared in The Chemical Intelligencer magazine published 1995 2000 Excerpts of these Editor s choice chapters chronicle the culture and history of chemistry featuring great chemists and discoverers Contributors from among the best known authors of the chemistry community including numerous Nobel laureates Features behind the scenes stories about pivotal discoveries intricacies of laboratory life and interactions among scientists favorite recipes of renowned researchers life histories and anecdotes Chapters detail the human side of science but also present scientific information communicated in an easy to perceive and entertaining way This unique book is not only aimed at chemists but individuals who are interested in the cultural aspects of our science

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