



SSPC: The Society for Protective Coatings Technology Guide 15

Field Methods for Extraction and Analysis of Soluble Salts on Steel and Other Nonporous Substrates

1. Scope

1.1 This Guide describes the most commonly used field methods for the extraction and analysis of soluble salts on steel and other nonporous substrates. Laboratory methods are only included for situations where more complete salt extractions are desired through rigorous laboratory retrieval and analysis protocols.

2. Description and Use

2.1 INTRODUCTION: Coatings applied on surfaces contaminated with soluble salts exceeding a certain concentration exhibit diminished performance. Likewise, self-passivating metals (not intended for coating such as stainless steel, aluminum and copper) can be compromised from elevated concentrations of residual soluble salts in corrosive environments. Soluble salt testing involves two basic steps: 1) extraction of salts from the surface into a solution and 2) the analysis of the solution. This Guide is intended to assist the user in selecting specific procedures for extracting and performing qualitative and/or quantitative soluble salts from steel and nonporous surfaces. It includes field methods for measuring total conductivity (fully automated or multi-step) and specific ions. It also offers guidance for destructive laboratory soluble salt extraction and testing. An overview of available field and laboratory techniques, methods of salt extraction and surface concentration calculations are provided in Tables 1 and 2 respectively. A discussion and sources on the efficiency of salt removal (also called extraction) from the metal surfaces of interest are provided in Appendix C.

2.1.1 Abbreviations:

Cl ⁻	chloride ion
cm ³	cubic centimeters
cm ²	square centimeters
Fe ⁺²	ferrous ion
μS	microSiemens
μS/cm	microSiemens per centimeter
mg	milligram
ml	milliliter
ppm	parts per million
NaCl	sodium chloride
NO ₃ ⁻	nitrate ion
SO ₄ ⁻²	sulfate ion

2.2 SALT EXTRACTION SOLUTIONS

All of the reagents used to extract salts from metal surfaces in this document fall into two categories: reagent water and proprietary aqueous solutions. They will be designated herein as reagent water or proprietary extraction solution.

2.2.1 Reagent Water: Reagent water used for salt extraction should have a maximum conductivity of 5 μS/cm. Distilled water may be purchased at grocery stores but verification of the conductivity is recommended. Alternatively, a portable demineralizer may be used to make reagent water on site. Pour tap water into the plastic bottle, attach the demineralizer cartridge in the direction indicated, invert, and squeeze out the desired amount of water (for many of the tests described below, at least 25 ml will be required). The cartridge can be used until the blue color turns brown, as indicated on the side of the cartridge. Once this occurs, replace the cartridge. Each cartridge should deionize approximately 3000 ml of water.

2.2.2 Proprietary Solutions: Proprietary extraction solutions may be included with some commercial test kits. These solutions should only be used for the soluble salts described in the kit instructions. Proprietary solutions are not suitable for measuring conductivity of extracted solutions.

3. Referenced Standards

The standards listed below are updated on a periodic basis. The user of this guide should consult the most recent revision.

3.1 SSPC STANDARDS AND JOINT STANDARDS:

SP S1NACE No. 1 White Metal Blast Cleaning

3.2 ASTM INTERNATIONAL STANDARDS*

ASTM D4327	Standard Test Method for Anions in Water by Chemically Suppressed Ion Chromatography
ASTM D512	Standard Test Methods for Chloride Ion in Water

* ASTM International, 100 Barr Harbor Drive, 2000, Conshohocken, PA 19380-2899. For referenced ASTM standards, visit the ASTM website, <http://www.astm.org>, or contact ASTM Customer Service at astm@astm.org. For Annual Book of ASTM Standards volume information, refer to the standards Document Summary page on the ASTM website.

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1. Understanding the eBook Sspc Technology Guide 15 Method A2
 - The Rise of Digital Reading Sspc Technology Guide 15 Method A2
 - Advantages of eBooks Over Traditional Books
2. Identifying Sspc Technology Guide 15 Method A2
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sspc Technology Guide 15 Method A2
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sspc Technology Guide 15 Method A2
 - Personalized Recommendations
 - Sspc Technology Guide 15 Method A2 User Reviews and Ratings
 - Sspc Technology Guide 15 Method A2 and Bestseller Lists
5. Accessing Sspc Technology Guide 15 Method A2 Free and Paid eBooks
 - Sspc Technology Guide 15 Method A2 Public Domain eBooks
 - Sspc Technology Guide 15 Method A2 eBook Subscription Services
 - Sspc Technology Guide 15 Method A2 Budget-Friendly Options

6. Navigating Sspc Technology Guide 15 Method A2 eBook Formats
 - ePub, PDF, MOBI, and More
 - Sspc Technology Guide 15 Method A2 Compatibility with Devices
 - Sspc Technology Guide 15 Method A2 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sspc Technology Guide 15 Method A2
 - Highlighting and Note-Taking Sspc Technology Guide 15 Method A2
 - Interactive Elements Sspc Technology Guide 15 Method A2
8. Staying Engaged with Sspc Technology Guide 15 Method A2
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sspc Technology Guide 15 Method A2
9. Balancing eBooks and Physical Books Sspc Technology Guide 15 Method A2
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sspc Technology Guide 15 Method A2
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sspc Technology Guide 15 Method A2
 - Setting Reading Goals Sspc Technology Guide 15 Method A2
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sspc Technology Guide 15 Method A2
 - Fact-Checking eBook Content of Sspc Technology Guide 15 Method A2
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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