

Reporting Category 1: Matter and the Periodic Table (12 questions)

C.4A	RS	differentiate between physical and chemical changes and properties
C.4B	SS	identify extensive and intensive properties
C.4C	SS	compare solids, liquids, and gases in terms of compressibility, structure, shape, and volume
C.4D	RS	classify matter as pure substances or mixtures through investigation of their properties
C.5A	SS	explain the use of chemical and physical properties in the historical development of the Periodic Table
C.5B	RS	use the Periodic Table to identify and explain the properties of chemical families, including alkali metals, alkaline earth metals, halogens, noble gases, and transition metals
C.5C	RS	use the Periodic Table to identify and explain periodic trends, including atomic and ionic radii, electronegativity, and ionization energy

Reporting Category 2: Atomic Structure and Nuclear Chemistry (9 questions)

C.6A	SS	understand the experimental design and conclusions used in the development of modern atomic theory, including Dalton's Postulates, Thomson's discovery of electron properties, Rutherford's nuclear atom, and Bohr's nuclear atom
C.6B	SS	understand the electromagnetic spectrum and the mathematical relationships between energy, frequency, and wavelength of light
C.6C	SS	calculate the wavelength, frequency, and energy of light using Planck's constant and the speed of light
C.6D	SS	use isotopic composition to calculate average atomic mass of an element
C.6E	RS	express the arrangement of electrons in atoms through electron configurations and Lewis valence electron dot structures
C.12A	SS	describe the characteristics of alpha, beta, and gamma radiation
C.12B	RS	describe radioactive decay process in terms of balanced nuclear equations
C.12C	SS	compare fission and fusion reactions

Reporting Category 3: Bonding and Chemical Reactions (14 questions)

C.7A	RS	name ionic compounds containing main group or transition metals, covalent compounds, acids, and bases, using International Union of Pure and Applied Chemistry (IUPAC) nomenclature rules
C.7B	RS	write the chemical formulas of common polyatomic ions, ionic compounds containing main group or transition metals, covalent compounds, acids, and bases
C.7C	RS	construct electron dot formulas to illustrate ionic and covalent bonds
C.7D	SS	describe the nature of metallic bonding and apply the theory to explain metallic properties such as thermal and electrical conductivity, malleability, and ductility
C.7E	SS	predict molecular structure for molecules with linear, trigonal planar, or tetrahedral electron pair geometries using Valence Shell Electron Pair Repulsion (VSEPR) theory
C.8A	SS	define and use the concept of a mole
C.8B	RS	use the mole concept to calculate the number of atoms, ions, or molecules in a sample of material
C.8C	SS	calculate percent composition and empirical and molecular formulas
C.8D	RS	use the law of conservation of mass to write and balance chemical equations
C.8E	SS	perform stoichiometric calculations, including determination of mass relationships between reactants and products, calculation of limiting reagents, and percent yield

Reporting Category 4: Gases and Thermochemistry (8 questions)

C.9A	RS	describe and calculate the relations between volume, pressure, number of moles, and temperature for an ideal gas as described by Boyle's law, Charles' law, Avogadro's law, Dalton's law of partial pressure, and the ideal gas law
C.9B	SS	perform stoichiometric calculations, including determination of mass and volume relationships between reactants and products for reactions involving gases
C.9C	SS	describe the postulates of kinetic molecular theory
C.11A	SS	understand energy and its forms, including kinetic, potential, chemical, and thermal energies
C.11B	SS	understand the law of conservation of energy and the processes of heat transfer
C.11C	RS	use thermochemical equations to calculate energy changes that occur in chemical reactions and classify reactions as exothermic or endothermic
C.11D	SS	perform calculations involving heat, mass, temperature changes, and specific heat
C.11E	SS	use calorimetry to calculate the heat of a chemical process

Staar Chemistry 2013

**Karin Andersson, Selma Brynolf, Fredrik
Lindgren, Magda Wilewska-Bien**

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Organoselenium Chemistry Brindaban C. Ranu, Bubun Banerjee, 2020-04-20 Organoselenium Chemistry is a unique resource in this branch of organic organometallic chemistry. The authors give an overview of synthesis strategies, introduce bioactive and environmentally friendly organoselenium compounds, and discuss their applications from organic synthesis to the clinic. *Super Master Persiapan AKM & SK dan Pendalaman Materi US/USP SMA/MA Kelas X SAINTEK* Tim Quantum Research, 2021-01-18 Asesmen Kompetensi Minimum Survei Karakter AKM SK merupakan salah satu kebijakan untuk merealisasikan Program Merdeka Belajar AKM SK. Ini merupakan pengganti Ujian Nasional UN yang selama ini sudah diselenggarakan. Adapun pelaksanaannya adalah di Kelas XI. Hal ini bertujuan supaya sekolah masih mempunyai waktu satu tahun di Kelas XII untuk memperbaiki apabila hasilnya kurang baik. Di samping itu, pemerintah mengembalikan proses ujian di kelas XII ke sekolah masing-masing sehingga dikenal dengan istilah Ujian Sekolah US atau Ujian Satuan Pendidikan USP. AKM SK bentuknya tidak hanya tes untuk menjawab soal, tetapi bisa juga berupa penugasan baik berupa Tugas Proyek atau Tugas Portofolio. Sementara itu, US USP masih berupa tes menjawab soal. Soal-soal yang akan diujikan digunakan untuk mengukur kemampuan siswa mulai dari jenjang C1 sampai C6. Bentuk soal tipe C1-C6 bisa Anda dapatkan di dalam buku ini. Oleh karena itu, buku ini bisa dijadikan solusi untuk menghilangkan kecemasan pada saat Anda akan menghadapi berbagai jenis ujian Matematika, Fisika, Kimia, Biologi, Bahasa Indonesia, dan Bahasa Inggris. Pembahasan di dalam buku ini difokuskan kepada penguasaan konsep. Melalui buku ini, Anda akan lebih mudah memahami konsep keenam mata pelajaran melalui intisari materi yang disajikan dan langsung diimplementasikan pada pemecahan masalah dalam bentuk soal dan pembahasan. Untuk mengukur penguasaan konsep untuk setiap sub pokok bahasan, Anda dapat mengerjakan Soal Pengetahuan dan Pemahaman serta Soal Aplikasi dan Penalaran. Adapun penguasaan konsep keseluruhan bab dapat Anda ukur dengan mengerjakan soal-soal Review. Buku ini juga dapat digunakan oleh guru sebagai referensi dalam penyusunan soal-soal Asesmen Kompetensi Minimum AKM Ujian Sekolah US atau Ujian Satuan Pendidikan USP dan ujian-ujian lainnya. Selamat Berlatih dan Berjuang Menjadi yang Terbaik. **Noncovalent Forces** Steve Scheiner, 2015-04-16 Computational methods and in particular quantum chemistry have taken the lead in our growing understanding of noncovalent forces as well as in their categorization. This volume describes the current state of the art in terms of what we now know and the current questions requiring answers in the future. Topics range from very strong ionic to very weak CH interactions. In the intermediate regime, forces to be considered are H bonds, particularly CH...O and OH...metal, halogen, chalcogen, pnictogen, and tetrel bonds, aromatic stacking, dihydrogen bonds, and those involving radicals. Applications include drug development and predictions of crystal structure. **The Politics of African-American Education** Kenneth J. Meier, Amanda Rutherford, 2016-08-24 The book provides a comprehensive assessment of current African American education policy and its politics. SWYK on STAAR Science Gr. 8, Student Workbook Show What You Know Publishing, 2013-03-01 Assess student

knowledge of the Texas Essential Knowledge and Skills TEKS for Science with two full length Assessments for each subject Questions provide students with the necessary practice needed to achieve academic success on STAAR Chapters on test taking strategies and test anxiety build students confidence and test taking skills Glossaries familiarize students with vocabulary terms and concepts found on state proficiency tests Answers are provided in the Parent Teacher Edition only

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