

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for

8th grade STAAR Test



but would work for other standardized tests

Crackables

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 75.1 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

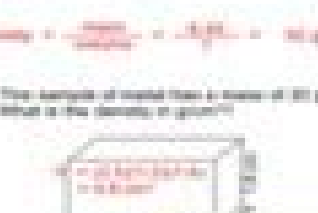
$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{75.1}{14.2} = 5.3 \text{ g/cm}^3$$

2. A rock is 1 cm³. What is the density of the rock in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{4.25}{1} = 4.25 \text{ g/cm}^3$$

3. The sample of liquid has a mass of 25 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{10} = 2.5 \text{ g/cm}^3$$

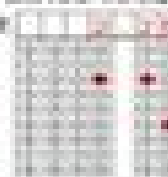
4. A rectangular block of metal has a mass of 20 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{20}{10} = 2 \text{ g/cm}^3$$

5. Several glass marbles and 25 mL of water are shown in the beakers. Be sure to use the correct glassware to measure the volume of the water.

1.



2.



3.



Design
81
STA

Staar Science Problems Griddable

Michael Rajnik



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