

STAAR Science Tutorial 07

TEK 6.5C: Elements & Compounds

TEK 6.5C: Differentiate between elements and compounds on the most basic level.

Elements & Atoms

- An **element** is a substance that cannot be separated into simpler substances by physical or chemical means. An element is already in its simplest form.
- The smallest piece of an element that still has the properties of that element is called an **atom**.
- An element is a pure substance, containing only one kind of atom. For example, the element gold is made only of gold atoms.
- Examples of elements include oxygen, potassium, carbon, hydrogen, sodium, and iron. The Periodic Table of Elements is a list of all the elements that have been discovered and named.
- Elements are represented on the Periodic Table by a one or two letter **symbol**. If it is a one letter symbol, it is always capitalized (such as S for sulfur). If it is a two letter symbol, the first letter is always capitalized and the second letter is always lowercase (such as He for helium).
- Most symbols look like abbreviations of the English element name. But some element symbols are based on the original Latin or Greek name for the element, such as gold (Au), silver (Ag), iron (Fe), sodium (Na), potassium (K) and mercury (Hg).

Compounds

- A **compound** is a substance that is made by chemically bonding two or more elements in a set ratio. For example, sodium (Na) and chlorine (Cl) in a 1:1 ratio makes the compound sodium chloride, commonly known as table salt. The symbols of the elements in a compound can be listed together to describe the chemical formula of the compound. Sodium chloride has the chemical formula "NaCl".
- Compounds have different chemical and physical properties than the elements from which they are made.
- For example, while sodium is a reactive metal and chlorine is a poisonous gas, the resulting white powder is neither a metal nor a gas, and is non-poisonous.

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Table of Contents Staar Science Tutorial 07 Answer Sheet

1. Understanding the eBook Staar Science Tutorial 07 Answer Sheet
 - The Rise of Digital Reading Staar Science Tutorial 07 Answer Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Staar Science Tutorial 07 Answer Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Staar Science Tutorial 07 Answer Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Staar Science Tutorial 07 Answer Sheet
 - Personalized Recommendations
 - Staar Science Tutorial 07 Answer Sheet User Reviews and Ratings
 - Staar Science Tutorial 07 Answer Sheet and Bestseller Lists
5. Accessing Staar Science Tutorial 07 Answer Sheet Free and Paid eBooks
 - Staar Science Tutorial 07 Answer Sheet Public Domain eBooks

- Staar Science Tutorial 07 Answer Sheet eBook Subscription Services
- Staar Science Tutorial 07 Answer Sheet Budget-Friendly Options
- 6. Navigating Staar Science Tutorial 07 Answer Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Staar Science Tutorial 07 Answer Sheet Compatibility with Devices
 - Staar Science Tutorial 07 Answer Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Staar Science Tutorial 07 Answer Sheet
 - Highlighting and Note-Taking Staar Science Tutorial 07 Answer Sheet
 - Interactive Elements Staar Science Tutorial 07 Answer Sheet
- 8. Staying Engaged with Staar Science Tutorial 07 Answer Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Staar Science Tutorial 07 Answer Sheet
- 9. Balancing eBooks and Physical Books Staar Science Tutorial 07 Answer Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Staar Science Tutorial 07 Answer Sheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Staar Science Tutorial 07 Answer Sheet
 - Setting Reading Goals Staar Science Tutorial 07 Answer Sheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Staar Science Tutorial 07 Answer Sheet
 - Fact-Checking eBook Content of Staar Science Tutorial 07 Answer Sheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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