

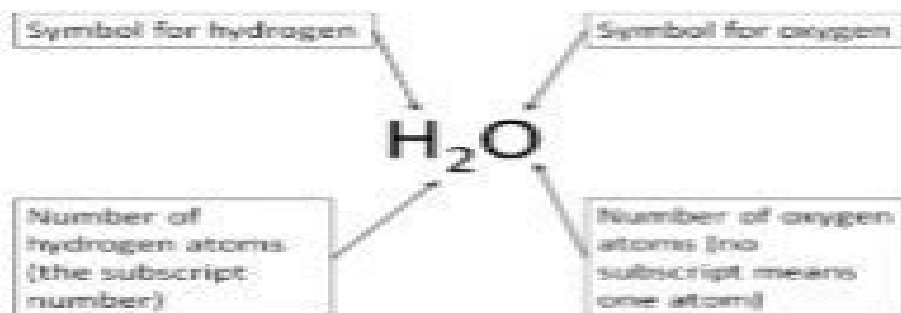
## **STAAR Science Tutorial 02**

### **TEK 8.5D: Chemical Formulas**

**TEK 8.5D:** *Recognize that chemical formulas are used to identify substances and determine the number of atoms of each element in chemical formulas containing subscripts.*

#### **Chemical Formulas**

- A **chemical formula** is a description of the number and kind of atoms found in a single molecule of a substance, using the symbol for each element and a subscript number to state the number of atoms of the element to the left of the subscript.
- Scientists use chemical formulas such as NaCl instead of common names (table salt) or chemical names (sodium chloride) because it is shorter, more accurate, and universally understood.
- Examples of chemical formulas include  $\text{H}_2\text{O}$  (water),  $\text{CO}_2$  (carbon dioxide),  $\text{HC}_2\text{H}_3\text{O}_2$  (vinegar),  $\text{NaHCO}_3$  (baking soda),  $\text{NH}_3$  (ammonia), and  $\text{C}_6\text{H}_{12}\text{O}_6$  (glucose).



- If asked how many different elements are in a chemical formula, remember that each element symbol starts with a capital letter. But be careful to not count an element twice if repeated in the formula. In  $\text{HC}_2\text{H}_3\text{O}_2$ , there are only three elements, carbon, hydrogen and oxygen, even though there are four capital letters. (The H symbol is repeated twice in the formula.)

# Staar Science Tutorial Tek 8 5d Chemical Formulas

**LL Leslie**

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