






8.2 The Nature of Covalent Bonding > Key Concepts

-  In covalent bonds, e⁻ sharing occurs so that atoms attain the configurations of noble gases.
-  In a coordinate covalent bond, the shared e⁻'s come from a single atom.
-  The octet rule is not satisfied in molecules with an odd # of e⁻ & in molecules where an atom has less, or more, than a complete octet of valence e⁻.
-  A large bond dissociation energy corresponds to a strong covalent bond.
-  In ozone, the bonding of oxygen atoms is a hybrid of the extremes represented by the resonance forms.

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M Walker



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