

Calculating pH of Acids – Practice

Instructions: Use the equations for pH to complete the problems below.

- 1) What is the pH of a 0.0266 M hydrobromic acid (HBr) solution?

$$\text{pH} = -\log[\text{H}^+] \rightarrow \text{pH} = -\log[0.0266] \rightarrow \boxed{\text{pH} = 1.58}$$

- 2) Calculate the hydrogen ion concentration of a solution with a pH of 6.3.

$$[\text{H}^+] = 10^{-\text{pH}} \rightarrow [\text{H}^+] = 10^{-6.3} \rightarrow \boxed{[\text{H}^+] = 5.01 \times 10^{-7}}$$

- 3) Calculate the pH of a 0.004 M hydrofluoric acid (HF) solution?

$$\text{pH} = -\log[\text{H}^+] \rightarrow \text{pH} = -\log[0.004] \rightarrow \boxed{\text{pH} = 2.40}$$

- 4) What is the hydrogen ion concentration of an acid with a pH of 2.21?

$$[\text{H}^+] = 10^{-\text{pH}} \rightarrow [\text{H}^+] = 10^{-2.21} \rightarrow \boxed{[\text{H}^+] = 6.17 \times 10^{-3}}$$

- 5) Calculate the pH of a 1.14 L solution that contains 0.19 moles of HCl.

$$[\text{H}^+] = \frac{\text{moles of solute}}{\text{liters of solvent}} \rightarrow \frac{0.19 \text{ mol}}{1.14 \text{ L}} \rightarrow 0.167 \text{ M}$$

$$\text{pH} = -\log[\text{H}^+] \rightarrow \text{pH} = -\log[0.167] \rightarrow \boxed{\text{pH} = 0.78}$$

Ph Practice Problems With Answers

AW Rasmussen



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Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori,Richard H.

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Accurately calculating medication dosages is a critical element in pharmaceutical care that directly affects optimal patient outcomes Unfortunately medication dosage errors happen in pharmacies in hospitals or even at home or in homecare settings everyday In extreme cases even minor dosage errors can have dire consequences Careful calculations are essential to providing optimal medical and pharmaceutical care Essential Math and Calculations for Pharmacy Technicians fills the need for a basic reference that students and professionals can use to help them understand and perform accurate calculations Organized in a natural progression from the basic to the complex the book includes Roman and Arabic Numerals Fractions and decimals Ratios proportions and percentages Systems of measurement including household conversions Interpretation of medication orders Isotonicity pH buffers and reconstitutions Intravenous flow rates Insulin and Heparin products Pediatric dosage Business math Packed with numerous solved examples and practice problems the book presents the math in a step by step style that allows readers to quickly grasp concepts The authors explain the fundamentals simply and clearly and include ample practice problems that help readers become proficient The focus on critical thinking real life problem scenarios and the self test format make Essential Math and Calculations for Pharmacy Technicians an indispensable learning tool

Survival Guide to General Chemistry Patrick E. McMahon,Rosemary McMahon,Bohdan

Khomtchouk,2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the

major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

Mathematics Manual for Water and Wastewater Treatment Plant Operators: Wastewater Treatment Operations Frank R.

Spellman,2023-07-31 To properly operate a waterworks or wastewater treatment plant and to pass the examination for a waterworks wastewater operator s license it is necessary to know how to perform certain calculations All operators at all levels of licensure need a basic understanding of arithmetic and problem solving techniques to solve the problems they typically encounter in the workplace Hailed on its first publication as a masterly account written in an engaging highly readable user friendly style the fully updated Mathematics Manual for Water and Wastewater Treatment Plant Operators Wastewater Treatment Operations covers all the necessary computations used in wastewater treatment today It presents math operations that progressively advance to higher more practical applications including math operations that operators at the highest level of licensure would be expected to know and perform Features Provides a strong foundation based on theoretical math concepts which it then applies to solving practical problems for both water and wastewater operations Updated throughout and with several new practical problems added Provides illustrative examples for commonly used waterworks and wastewater treatment operations covering unit process operations found in today s treatment facilities

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