

Pharmaceutical Chemistry

QUESTIONNAIRE:

1. The water ion theory of acids and bases is attributed to:
A. Lewis
B. Bronsted
C. Arrhenius
D. Pauling
E. Pearson
2. This is radioactive particle that weights almost the same as helium atom.
A. x-ray
B. alpha ray
C. gamma ray
D. beta ray
E. ultraviolet ray
3. This is a sulfur containing ion that does not react with sulfuric acid?
A. Sulfide
B. Sulfite
C. Sulphate
D. thiosulfate
E. all of the given
4. A mole of a substance:
A. Contains Avogadro's number particles
B. Weighs the same as its atomic weight or molecular weight expressed in grams
C. Contains the same number of particles as 12 grams of Carbon
D. A and B
E. All of the above
5. Which of the following properties of matter is dependent on the quantity of matter present?
A. Flammability
B. Volume
C. Boiling point
D. reactivity
E. density
6. An attraction of molecule by a magnetic field doe to completely paired electrons.
A. Diamagnetism
B. paramagnetism
C. dia-para magnestism
D. ferromagnetism
E. none of the given
7. Which of the following substances is pure?
A. 5% dextrose
B. Table sugar
C. Blood
D. rubbing alcohol
E. salt solution
8. An atom with an atomic number of 15 and a mass number of 29 has _____ neutrons
A. 14
B. 15
C. 29
D. 39
E. 54
9. Two electrons occupying the same orbital differ in their:
A. Principal quantum number
B. Azimuthal quantum number
C. Magnetic quantum number
D. spin quantum number
E. angular momentum quantum number
10. What particles account for most of the mass of an atom?
A. Proton
B. Electron
C. Neutron
D. a and b
E. a and c
11. Which of the following properties of elements decreases from left to right of the periodic table.
A. Atomic size
B. Electron affinity
C. Electronegativity
D. non-metallic property
E. ionization potential

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Chandresh Agrawal, Nandini Books



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