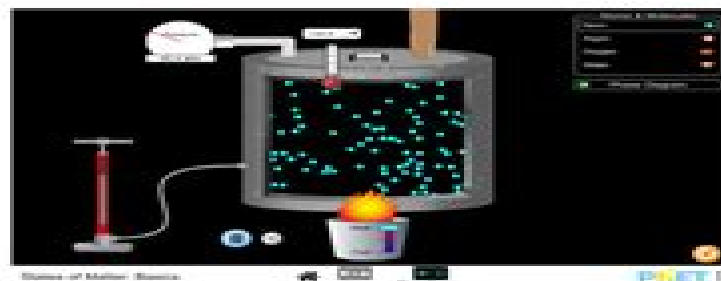


## Gas Laws Simulation Lab

- Go to [Phet.colorado.edu](http://Phet.colorado.edu)
- Search for the "States of Matter: Basics" simulation.
- Click on the "Phase Changes" icon.
- Select "Oxygen."



1. The simulation shows a quantity of oxygen molecules. What state of matter is this?
2. How do you know?  
Particles have a define shape
3. Heat the oxygen to about 150 K. What state of matter is this?
4. How do you know?  
because they are all spread out and have no define shape

### Experiment One: Constant volume.

- Leave the oxygen at 150 K. In the table below, record the pressure and temperature of the oxygen in the first row.
- Heat the oxygen to about 450 K.
- Record the new temperature and pressure in the second row of the table.
- Heat the oxygen to about 900 K.
- Record the new temperature and pressure in the third row of the table.

Temperature vs. pressure at constant volume	
Temperature (K)	Pressure (atm)
150	2.5
450	11.8
900	35.5

5. Write a statement describing how the temperature of a quantity of a gas is related to its pressure at constant volume.  
As the temperture rises, so does pressure

# Phet Gas Law Simulation Answers

**Mark Steyvers**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Phet Gas Law Simulation Answers:**

## Unveiling the Magic of Words: A Review of "**Phet Gas Law Simulation Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Phet Gas Law Simulation Answers**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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