

Gas Laws Computer Lab

Purpose

- 1.) To identify and describe the 4 properties of a gas sample: moles (n), volume (V), temperature (T), pressure (P)
- 2.) To determine the relationship between 2 variables at a time while keeping the other 2 variables constant
 - a. P and V (while keeping n and T constant)
 - b. V and T (while keeping n and P constant)
 - c. P and T (while keeping n and V constant)
 - d. V and n (while keeping P and T constant)

Definition of gas variables

A sample of gas in a closed container has 4 main characteristics:

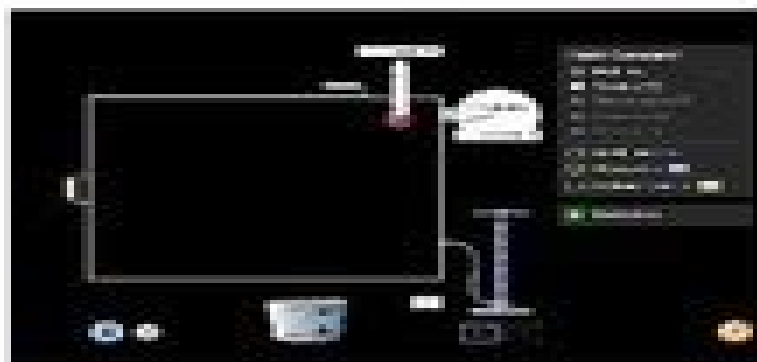
- a. Number of _____ of gas (n) – the quantity of gas particles present
- b. Volume of gas (V) – the amount of _____ the gas sample occupies
- c. _____ of gas (T) – the average kinetic energy of the gas particles (i.e. how fast they're moving)
- d. Pressure of gas (P) – the frequency and forcefulness of _____ between particles and container walls

Procedure

Part A – Set Up

- 1.) Google "phet gases molar", click the first search result, click PLAX button, double-click "Gases", or go to <https://phet.colorado.edu/en/simulations/where-phet/gases-molar>

Your screen should look like this:



- 2.) Open the "Particles" box. Make sure the like pump is purple color for heavy particles. Move the like pump handle to the highest position and push down to pump gas particles into the container. There are now 30 heavy gas particles in the container. In the "Particles" box, use the arrow buttons to make 100 heavy gas particles in container.

Phet Gas Law Simulation Lab Answers

Franklin S. Allaire, Jennifer E. Killham

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the authors' names.

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