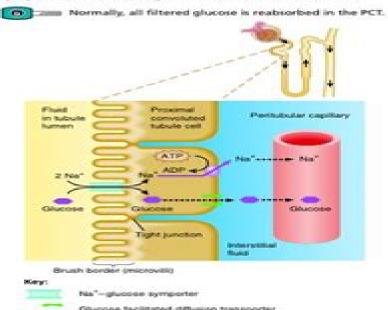
1084 CHAPTER 26 . THE URINARY SYSTEM

Figure 26.12 Reabsorption of glucose by Na⁺-glucose symporters in cells of the proximal convoluted tubule (PCT).



Kery:

Toa"—glucose symporter

Chucose facilitated diffusion transporter

Diffusion

Sodium-potassium pump

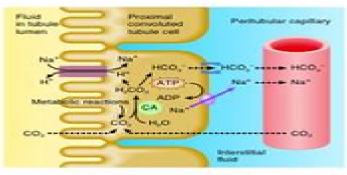
How does filtered glucose enter and leave a PCT cell?

cell in the PCT. Two Na⁺ and a molecule of glucose attach to the symporter protein, which carries them from the tubular fluid into the tubule cell. The glucose molecules then exit the basolateral membrane via facilitated diffusion and they diffuse into peritubular capillaries. Other Na⁺ symporters in the PCT reclaim filtered HPO₄⁺⁻ (phosphate) and SO₄⁺⁻ (sulfate) ions, all amino acids, and lactic acid in a similar way.

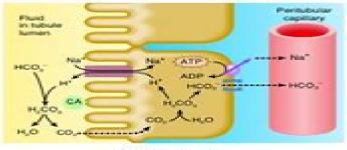
In another secondary active transport process, the Na*/H* antiportiers carry filtered Na* down its concentration gradient into a PCT cell as H* is moved from the cytosol into the lumen (Pigure 26.13a), causing Na* to be reabsorbed into blood and H* to be secreted into tubular fluid. PCT cells produce the H* needed to keep the antiporters running in the following way. Carbon dioxide (CO₂) diffuses from perimbular blood or tubular fluid or is produced by metabolic reactions within the cells. As also occurs in red blood cells (see Figure 23.25), the enzyme carbonic anhydruse (CA) (an-HI-drles) catalyzes the reaction of CO₂ with

Figure 25.13 Actions of Na*/H* antiporters in proximal convoluted tubule cells. (a) Reabsorption of sodium ions (Na*) and secretion of hydrogen ions (H*) via secondary active transport through the apical membrane; (b) reabsorption of bicarbonate ions (HCO₃**) via facilitated diffusion through the basolateral membrane. $CO_2 = carbon dioxide; H_2CO_3 = carbonic acid; CA = carbonic antivolrane.$

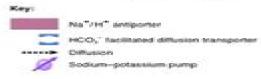
Na "/H" antiporters promote transcellular reabsorption of Na " and secretion of H".



(a) Na" reabsorption and H" secretion



(b) HCO," reabsorption



Which step in Na⁻ movement in part (a) is promoted by the electrochemical gradient?

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