

Name : _____ Date : _____



pH and pOH Worksheet



1. A solution has a hydrogen ion concentration of 2.8×10^{-6} M. What is the pOH of its solution ?

2. A solution has a $[\text{OH}^-]$ of 5.8×10^{-7} . What is the pH of this solution ?

3. A 450 mL beaker is 0.00045 M HCl. What is the pH of this solution ?

4. A 320 mL solution contains 2.30 mg of NaOH. What is the pH of this solution ?

5. Complete the following table.

pH	$[\text{H}^+]$	$[\text{OH}^-]$	pOH
3.5			
	5.8×10^{-7}		
		4.2×10^{-2}	
			8.2
	4.2×10^{-5}		
			2.4
10.1			
		7.2×10^{-3}	

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K Morrison



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