

Acid-Base Solutions – PhET Simulator

Instructions: Open the Acid-Base Solutions simulator on the PhET website/app. Choose “Introduction.”

- pH Test Strips change color when they come in contact with a solution.
- The color of the strip is compared to a pH Color Key to determine the pH.



- 1) Select the pH Test Strip from the Tools menu. Dip it into water. What is the pH? 7
- 2) Use pH test strips to test the Strong Acid, Weak Acid, Strong Base, and Weak Base. Draw the symbol for each solution in the appropriate location on the pH Color Key below:

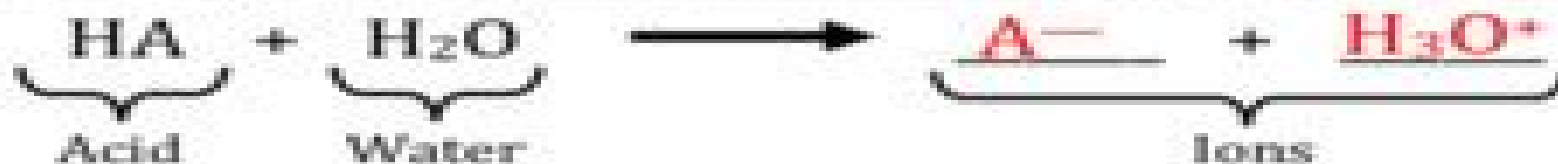


Draw each symbol under the correct pH

SA	WA	SB	WB
Strong Acid	Weak Acid	Strong Base	Weak Base

- When acids and bases dissolve in water, their ions dissociate (separate).
- Acids dissociate by releasing H⁺ ions. These H⁺ ions bond to H₂O to form H₃O⁺.

- 3) Choose the Strong Acid. Complete the chemical formula for dissociation of a strong acid:



Instructions: Under “Views,” choose the Graph option.

- a) What is the concentration of ions (A⁻ and H₃O⁺) in the Strong Acid? 1.00×10^{-2}
- b) What is the concentration of ions (A⁻ and H₃O⁺) in the Weak Acid? 3.16×10^{-5}
- c) Which type of acid (Strong or Weak) produces more ions? Strong Acid

Phet Lab Answer Key For Ph Scale

Thomas Griffiths



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