

Naming Ionic Compounds

What are the structural units that make up ionic compounds and how are they named?

Why?

When working in chemistry, it is often convenient to write a chemical in symbols. For example we might write down the substance table salt as NaCl . In talking about chemistry however, it is a bit tacky to say "en-ay see-ell" when we want to refer to a substance. Also, in formal writing we should use the name of the compound rather than its symbols. Therefore we need to learn how to say the proper names of ionic substances.

Model 1 – Ion Charges for Selected Elements

| | | | | | | | | | | | | | | | | |
|-------------|-----------------|------------------|--------------------------------------|--------------------------------------|-------------------------------------|---|--|--|------------------|--------------------------------------|--------------------------------------|-----------------|-----------------|-----------------|--|--|
| 1 | H ⁺ | | | | | | | | | | | | | | | |
| 2 | Li ⁺ | Be ²⁺ | Transition elements | | | | | | Al ³⁺ | | N ³⁻ | O ²⁻ | F ⁻ | | | |
| 3 | Na ⁺ | Mg ²⁺ | | | | | | | | | P ³⁻ | S ²⁻ | Cl ⁻ | | | |
| 4 | K ⁺ | Ca ²⁺ | Fe ²⁺ Fe ³⁺ | Ni ²⁺ Ni ³⁺ | Cu ⁺ Cu ²⁺ | Zn ²⁺ | | | | | | | | Br ⁻ | | |
| 5 | Rb ⁺ | Sr ²⁺ | | | Ag ¹⁺ | | | | | | Sn ²⁺ Sn ⁴⁺ | | | I ⁻ | | |
| 6 | | Ba ²⁺ | | | | Hg ₂ ²⁺ Hg ²⁺ | | | | Pb ²⁺ Pb ⁴⁺ | | | | | | |
| ← Cations → | | | | | | | | | | | | | | | | |

1. Based on the information in Model 1:

- a. Identify three elements that form only one cation. \leftarrow Positive Ion
 Na^+ , Sr^{2+} , Al^{3+} (Any box with one element works.)
- b. Identify three elements that form only one anion. \leftarrow Negative Ion
 P^{3-} , O^{2-} , Br^- (Any box with one element works.)
- c. Identify three elements that form more than one cation.
 Fe^{+2} and Fe^{+3} , Ni^{+2} and Ni^{+3} , Cu^{+1} and Cu^{+2}
- d. In what region of the periodic table are these "multiple ion" elements usually located?
Transition elements or "d" block

2. Consider the ions of potassium (K) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of potassium (K) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.
- $2K^+$ and S^{2-} or K_2S
3. Consider the ions of iron (Fe) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of iron (Fe) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.



Traditional Names For Ionic Compounds Answer Key

Amar Bhutani



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Bold Ventures Raizen, 2012-12-06 This book presents comprehensive results from case studies of five innovations in science education that have much to offer toward understanding current reforms in this field Each chapter tells the story of a case in rich detail with extensive documentation and in the voices of many of the participants the innovators the teachers the students Similarly Volume 3 of Bold Ventures pre sents the results from case studies of five innovations in mathematics education Volume 1 provides a cross case analysis of all eight innovations Many U S readers certainly will be very familiar with the name of at least one if not all of the science innovations discussed in this volume for example Project 2061 and probably with their general substance Much of the education community s familiarity with these arises from the projects own dissemination efforts The research reported in this volume however is one of the few detailed studies of these innovations undertaken by researchers outside the projects them selves Each of the five studies was a large scale effort involving teams of researchers over three years These teams analyzed many documents attended numerous critical project meetings visited multiple sites conducted dozens of individual interviews The team leaders Atkin Huberman Rowe having spent much time with science education over long careers looked at these innovations through many lenses It was a daunting task for each team to sift through the mountains of detail in order to bring the most compelling themes to the surface **Investigating**

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