



# Trane Tracer Sc Manual

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web aug 23 2019 these are called polyatomic ions table 2 lists the formulas charges and names of some common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions

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web polyatomic ions are charged groups of atoms an example is ammonium ion  $\text{NH}_4^+$  it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is  $\text{NH}_4\text{Cl}$  and ammonium sulfide is  $\text{NH}_4\text{S}$

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web polyatomic ions have defined formulas names and charges that cannot be modified in any way table 3 4 1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example  $\text{NO}_3^-$  is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion  $\text{NO}_3^-$  contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written

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