

Key

Is it a physical or chemical change?

Example	Chemical Equation	Physical or chemical change?
Melting ice	$\text{H}_2\text{O (solid)} \rightarrow \text{H}_2\text{O (liquid)}$	Physical
Iron-Rusting	$4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$	Chemical
Breaking a pencil	$\text{Pencil (solid)} \rightarrow \text{Pencil (shavings)}$	Physical
Boiling water	$\text{H}_2\text{O (liquid)} \rightarrow \text{H}_2\text{O (gas)}$	Physical
Burning Paper	$\text{Paper} + \text{O}_2 \rightarrow \text{Ash} + \text{CO}_2 + \text{H}_2\text{O}$	Chemical
Baking Soda & Vinegar	$\text{NaHCO}_3 + \text{CH}_3\text{COOH} \rightarrow \text{NaCH}_3\text{COO} + \text{H}_2\text{O} + \text{CO}_2$	Chemical
Dissolving Salt in Water	$\text{NaCl (solid)} \rightarrow \text{NaCl (aq)}$	Physical
Photosynthesis	$6\text{CO}_2 + 6\text{H}_2\text{O} \xrightarrow{\text{light}} \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$	Chemical

Explain the difference between a chemical change and a physical change. Use the examples provided to support your explanation.

A chemical change is a process in which a substance is transformed into one or more new substances. This transformation is usually indicated by a change in color, the formation of a precipitate, the release of a gas, or the formation of a new substance with different properties. Examples of chemical changes include burning, rusting, and cooking.

A physical change is a process in which a substance is transformed into one or more new substances, but the chemical composition of the substance remains the same. Examples of physical changes include melting, boiling, and dissolving.

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Explain the difference between a chemical change and a physical change. Use the examples provided to support your explanation.

Test students' understanding

Physical And Chemical Changes Review And Reinforce

J Elliott



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