

Photoelectric Effect

When light of a suitable frequency is allowed to incident on a metal, ejection of electrons take place. This phenomenon is known as photo electric effect.

Einstein theory of photo electric effect.

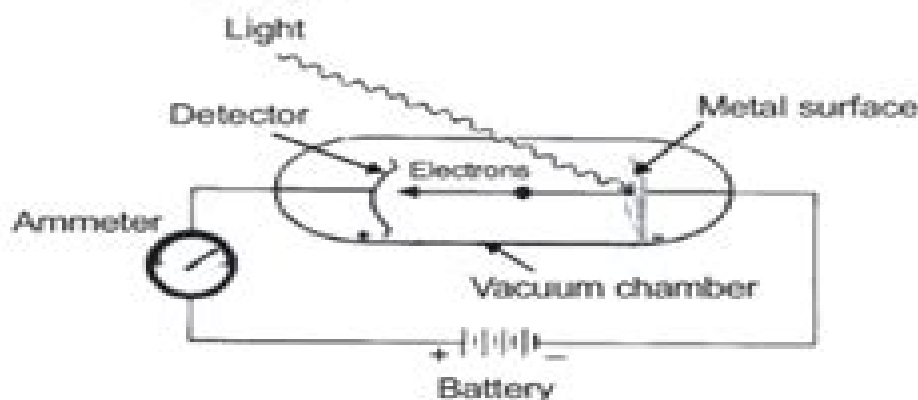


Figure : Equipment for studying the photoelectric effect

Light of a particular frequency strikes a clean metal surface inside a vacuum chamber. Electrons are ejected from the metal and are counted by a detector that measures their kinetic energy.

Main Point

1. One photon of suitable frequency can eject one electron
2. Energy of the photon can be utilised in two ways :
 - (i) Its one part is used to just eject the electron, this part is known as work function. Work function can be calculated as

$$W = h\nu_0, \text{ where } \nu_0 \text{ is threshold frequency}$$

$$W = \frac{hc}{\lambda_0}, \lambda_0 \rightarrow \text{threshold wavelength}$$

Threshold frequency : Minimum frequency that is required to just eject electron.

- (ii) Remaining part of energy is used to provide kinetic energy.

So, Einstein equation is as given below

$$\underbrace{h\nu}_{\text{Incident photon}} = \underbrace{h\nu_0}_{\text{Work function}} + \underbrace{KE}_{\text{Kinetic energy}}$$

Photoelectric Effect Test Question

Oswaal Editorial Board



Photoelectric Effect Test Question:

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