

star X compare to the position 1.

- 1c. For light cannot escape, then escape velocity = $3 \times 10^8 \text{ ms}^{-1}$.

$$v = \sqrt{\frac{2GM}{R}} = \sqrt{\frac{2G(3.82 \times 10^6 M_\odot)}{R}}$$

$$3 \times 10^8 = \sqrt{\frac{2 \times 1.33 \times 10^{28} \times 3.82 \times 10^6}{R}}$$

$$\therefore R = 1.129 \times 10^{10} \text{ m} = 0.0752 \text{ AU}$$

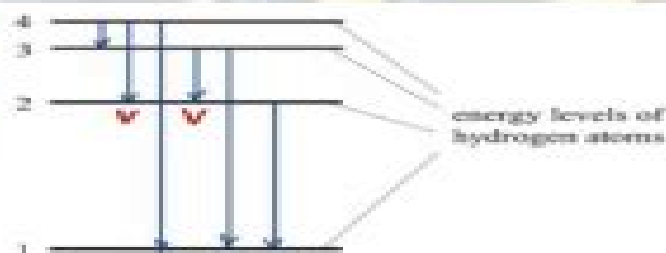
- 2a. Rutherford's model stated that the EM waves are radiated continuously as the electron has centripetal acceleration, the atom loses energy continuously. It would be unstable and collapse finally, but actually the atom is stable and does not collapse.

- 2bi. C

- 2bii. $\lambda = 364.6 \left(\frac{n^2}{n^2 - 2^2} \right) = 434 \text{ nm (violet)}$

- 2biii. The photon will be absorbed by the hydrogen atom. The electron of the hydrogen atom will then activate and escape as free electron. Then the hydrogen atom is said to be ionized.

- 2biv. There are 6 possible electron transitions.



- 3a. It is because the mass of A is larger than B, friction of A is larger than B. Thus more energy should be used to overcome friction.

Phy Dse Paper

Thomas Bond,Chris Hughes



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