

Nuclear Decay and Radioactivity

12. Identify the products resulting from alpha decay of the following radioisotopes.

(a) ${}^{238}\text{Pu}$ _____

(c) Lanthanum-138 _____

(b) ${}^{238}\text{Po}$ _____

(d) Americium-243 _____

(e) ${}^{241}\text{Fr}$ _____

(f) Protactinium-233 _____

(g) ${}^{238}\text{Pa}$ _____

(h) Bismuth-210 _____

(i) ${}^{234}\text{Ac}$ _____

(j) Radium-226 _____

13. Identify the products resulting from beta decay of the following radioisotopes.

(a) ${}^{137}\text{Ba}$ _____

(c) Barium-82 _____

(b) ${}^{23}\text{Li}$ _____

(d) Calcium-47 _____

(e) ${}^{49}\text{V}$ _____

(f) Phosphorus-32 _____

(g) ${}^{130}\text{I}$ _____

(h) Carbon-14 _____

(i) ${}^{129}\text{I}$ _____

(j) Mercury-203 _____

14. Radium-226 has a half-life of 1600 years. How much of a 2 kg sample will remain after:

(a) 1600 years? _____

(b) 3200 years? _____

(c) 4800 years? _____

(d) 6400 years? _____



Radioactivity And Nuclear Reactions Answers

John D. Cutnell, Kenneth W. Johnson



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