+ WS 6.2 - Nuclear Reactions

Complete each of the following nuclear reactions by determining the missing particle, then name that particle ("alpha particle" or "uranium-233", etc...) #1 is an example...

1. 14C -> 14N + 9e beta particle	symbol ⁴He+≥ or α	name eloha particle
	21.10.2 Of C	alpha particle (a helium nucleus)
2. + 9e> +0Ar	ρe or β	beta particle jan electroni
3. 252C1> + 196Mo + 41n	0er +1	positron (a positive electron?)
	in.	neutron
4. 9Be +> 12C + in	*p or 1H+1	proton (hydrogen-1 nucleus)
5. gLi + 12C> 12F +	2 H +1	deutron (hydrogen-2 nucleus)

- 7. 3He + 3He ---> 7Be +
- 8. 235U + 1n --> 144Cs + 90Rb +
- 9. --> 4He + 229Th _____
- 10. 39Si + -> 30P _____
- 11. 19K + 5n ---> 11Ar +
- 12. 24Mg + -> 22Na + a
- 13. 200Th + β --> 200Po + 2 2H+1 +
- 14. 246Pu + 1n --> 96Mo + + 51n (find the storate #at)
- 15. Write the reaction for a Rn-224 atom undergoing 1 beta decay: What is the daughter?
- 16. Write the reaction for a Po-218 atom undergoing 1 alpha decay: What is the daughter?
- 17. What particle is produced when a U-234 atom undergoes a series of two alpha decays and three beta decays?
- 18. Pb-210 undergoes a series of alpha and beta decays to end up eventually as Ir-198. How many alpha particles and how many beta particles were emitted in all? Hint: do alpha's first, then beta's. $\alpha = \beta = \beta$

Radioactivity And Nuclear Reactions Worksheet Answers

Hussin A.Rothana

Radioactivity And Nuclear Reactions Worksheet Answers:

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Table of Contents Radioactivity And Nuclear Reactions Worksheet Answers

- 1. Understanding the eBook Radioactivity And Nuclear Reactions Worksheet Answers
 - The Rise of Digital Reading Radioactivity And Nuclear Reactions Worksheet Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Radioactivity And Nuclear Reactions Worksheet Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - $\circ \ \ Determining \ Your \ Reading \ Goals$
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Radioactivity And Nuclear Reactions Worksheet Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Radioactivity And Nuclear Reactions Worksheet Answers
 - Personalized Recommendations

- Radioactivity And Nuclear Reactions Worksheet Answers User Reviews and Ratings
- Radioactivity And Nuclear Reactions Worksheet Answers and Bestseller Lists
- 5. Accessing Radioactivity And Nuclear Reactions Worksheet Answers Free and Paid eBooks
 - Radioactivity And Nuclear Reactions Worksheet Answers Public Domain eBooks
 - Radioactivity And Nuclear Reactions Worksheet Answers eBook Subscription Services
 - Radioactivity And Nuclear Reactions Worksheet Answers Budget-Friendly Options
- 6. Navigating Radioactivity And Nuclear Reactions Worksheet Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Radioactivity And Nuclear Reactions Worksheet Answers Compatibility with Devices
 - o Radioactivity And Nuclear Reactions Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Radioactivity And Nuclear Reactions Worksheet Answers
 - Highlighting and Note-Taking Radioactivity And Nuclear Reactions Worksheet Answers
 - Interactive Elements Radioactivity And Nuclear Reactions Worksheet Answers
- 8. Staying Engaged with Radioactivity And Nuclear Reactions Worksheet Answers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Radioactivity And Nuclear Reactions Worksheet Answers
- 9. Balancing eBooks and Physical Books Radioactivity And Nuclear Reactions Worksheet Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Radioactivity And Nuclear Reactions Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Radioactivity And Nuclear Reactions Worksheet Answers
 - Setting Reading Goals Radioactivity And Nuclear Reactions Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Radioactivity And Nuclear Reactions Worksheet Answers
 - Fact-Checking eBook Content of Radioactivity And Nuclear Reactions Worksheet Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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