

1.2. Identify the products resulting from alpha decay of the following radioisotopes.

- |                             |                            |
|-----------------------------|----------------------------|
| 040 $^{238}\text{Pu}$ _____ | 075 Lanthanum-138 _____    |
| 040 $^{239}\text{Po}$ _____ | 080 Bismuth-213 _____      |
| 040 $^{238}\text{Fr}$ _____ | 090 Protactinium-233 _____ |
| 040 $^{238}\text{Ra}$ _____ | 095 Lanthanum-138 _____    |
| 040 $^{238}\text{Ac}$ _____ | 100 Lead-186 _____         |

1.3. Identify the products resulting from beta decay of the following radioisotopes.

- |                             |                         |
|-----------------------------|-------------------------|
| 040 $^{238}\text{Pa}$ _____ | 075 Bismuth-185 _____   |
| 040 $^{238}\text{Li}$ _____ | 080 Calcium-45 _____    |
| 040 $^{238}\text{F}$ _____  | 090 Phosphorus-32 _____ |
| 040 $^{238}\text{Sr}$ _____ | 095 Carbon-14 _____     |
| 040 $^{238}\text{Rb}$ _____ | 100 Potassium-40 _____  |

1.4. Protactinium-233 has a half-life of 27,000 years.  
How much of a 2 kg sample will remain after \_\_\_\_\_

- |                         |                         |
|-------------------------|-------------------------|
| 040 40,000 years? _____ | 040 34,000 years? _____ |
| 040 80,000 years? _____ | 040 40,000 years? _____ |

1.5. What are three ways that nuclear reactions are different to chemical reactions?

- 03 \_\_\_\_\_
- 040 \_\_\_\_\_
- 040 \_\_\_\_\_

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**Clemens Wendtner**



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