1. Name the following compounds as acids and circle which are weak acids:

a) HNO

b) H<sub>2</sub>SO<sub>4</sub>

(d) Hacos

(e) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

mittric a old

sulfuric acid hydrofluoric acid carbonic acid

a cetti o la cild

phosphoric acid

2. Write the formulas for the following bases:

Ca [OH]<sub>2</sub>

a) calcium hydroxide b) aluminum hydroxide c) ammonium hydroxide d) lithium hydroxide ALIOHIS

NHLOH.

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3. Complete the following table:

	Acids	Bases
Definition	A substance that increases the concentration of H" (or H <sub>2</sub> O") when added to water.	A substance that increases the concentration of OHT when added to water.
	or	or
	A substance that acts as a proton (H*) donor.	A substance that acts as a proton (H*) acceptor.
Electrolyte in aqueous solution?	H*	OHT
рН	less than 7	greater than 7
Taste	sour	bitter
Litmus turns	blue turns red	red turns blue
Which reacts with metals to form H <sub>2</sub> gas?	yes	no

#### Conjugate acid-base pairs:

4. Identify the acid, the base, the conjugate acid and the conjugate base in the reactions:

 $NH_{2}(g) + H_{2}O(l) \neq NH_{4}(aq) + OH(aq)$ base ac lid cont. acid cont. base

HCO3" (aq) + H2O (1) CO<sub>3</sub>2" (aq) + H<sub>2</sub>O" (aq) conj. acid and help base conf. base

5. An acid will form the conjugate \_base\_and a base will form the conjugate \_acid\_\_.

# **Test Acids And Bases Answers**

**Arturo Cuomo** 

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