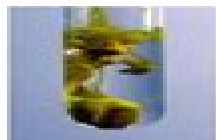


1. Which is the name of the kind of solid substance formed in this figure?



- | | |
|-------------------------------------|------------------------------------|
| <input type="radio"/> A precipitate | <input type="radio"/> B acid |
| <input type="radio"/> C aqueous | <input type="radio"/> D synethesis |

2. An element plus additional element reacts to form one product is an example of which type of chemical reaction?

- | | |
|--|---------------------------------------|
| <input type="radio"/> A single replacement | <input type="radio"/> B decomposition |
| <input type="radio"/> C synthesis | <input type="radio"/> D combustion |

3. Describes the number of molecules in a compound and is used to balance a chemical reaction.

- | | |
|-------------------------------------|-------------------------------------|
| <input type="radio"/> A SI unit | <input type="radio"/> B subscript |
| <input type="radio"/> C superscript | <input type="radio"/> D coefficient |

4. Which type of reaction takes place in the presence of oxygen and produces carbon dioxide and water?

- | | |
|--|--|
| <input type="radio"/> A decomposition | <input type="radio"/> B combustion |
| <input type="radio"/> C single replacement | <input type="radio"/> D double replacement |

5. Provides evidence that a chemical reaction has occurred.

- | | |
|--|------------------------------------|
| <input type="radio"/> A melting | <input type="radio"/> B bending |
| <input type="radio"/> C formation of a gas | <input type="radio"/> D dissolving |

6. What is the probable product of a double-replacement reaction?

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C Cleary



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