

4. **Practice:** Click **Reset**, turn off **Show equation**, and select **Polonium**. Write an equation for the alpha decay of polonium, and then use the Gizmo to check your answer.



What isotope remains after the alpha decay of polonium-212?

Pb-208

5. **Practice:** Click **Reset**, turn off **Show equation**, and select **Radium**. Write an equation for the alpha decay of radium, and then use the Gizmo to check your answer.



What dangerous gas is produced by the decay of radium-226?

Rn-222

6. **Practice:** Americium-241 is a radioactive isotope used in smoke detectors. Write an equation for the alpha decay of Americium-241 below.



7. **Analyze:** In each equation, how is the mass number on the left side of the arrow related to the sum of mass numbers on the right side of the arrow? Is this true for atomic numbers?

They are balanced, and atomic numbers are included in this.

8. **Summarize:** In general, how can you determine the mass number of the daughter product after alpha decay has taken place? How can you determine the atomic number?

The original mass number is decreased by 4. By deducting 2 from the original atomic number, you can calculate the new one.

9. **Think and discuss:** Helium is the second most abundant element in the universe, but it is rare in Earth's atmosphere. Most of the helium used to fill balloons and blimps must be extracted from Earth's crust. How do you think this helium formed?

The alpha particles released by radioactive isotopes are what create this helium.

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