

Score: 0 %

## Question 1.

✓ Correct

Which of the following components stores values in short-term memory?

- ☒ RAM
- ☐ Processor
- ☐ Flash drive
- ☐ Hard drive

## Explanation:

RAM stores values in short-term memory.

Hard drives store data in long-term memory.

Processors execute instructions stored in computer programs.

Flash drives are portable devices that store data in long-term memory.

## References:

-  3.3.1Internal Computing Components
-  3.3.2Internal Computing Component Facts
- q\_int\_intocomponents\_facts\_01.question.tex

## Question 2.

✓ Correct

Which hardware component processes data?

- ☐ SSD
- ☒ CPU
- ☐ HDD
- ☐ RAM

## Explanation:

The CPU, or Central Processing Unit, processes data.

RAM stores values in short-term memory.

HDD and SSD are two types of long-term memory storage.

## References:

-  3.3.2Internal Computing Component Facts
- q\_int\_intocomponents\_facts\_02.question.tex

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web a determine the reaction order for hci using calculations described in the background section show your work note that your answer will probably not be an even whole

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web the rate law for a particular reaction is rate  $k x^2$  in an experiment the initial rate of the reaction is determined to be 0.080 mol l s when the initial concentration of x is

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web a rate law shows how the rate of a chemical reaction depends on reactant concentration for a reaction such as  $a \rightarrow \text{products}$  the rate law generally has the form rate  $k a^n$

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web the reaction orders in a rate law describe the mathematical dependence of the rate on reactant concentrations referring to the generic rate law above the reaction is  $m$

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web each reactant in the rate limiting step of a reaction is assigned an order typically zeroth 1st or 2nd the order is the sum of all individual orders the rate of a reaction is

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web feb 13 2023 1 define reaction order use the following information to solve questions 2 and 3 given the rate law equation text rate  $k$  a 1 b 2 nonumber 2 determine

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web expert answer transcribed image text 2 determine the reaction orders and the rate law for the experiment determine the rate constant at room temperature 3 explain how the

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web the rate law or differential rate law relates the rate of a reaction to the concentration or pressure of the reactants the rate of a reaction is proportional to the concentration or

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web oct 27 2022 the rate law for the reaction  $\text{ce h}_2 \text{ g ce } 2\text{no g ce n}_2\text{o g ce h}_2\text{o g}$  nonumber has been experimentally determined to be rate

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web the absorbance of a particular wavelength of light by a solution 2 in this experiment you will conduct the reaction between solutions of potassium iodide and iron iii chloride

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web 1 use table 1 to determine the reaction order for hcl 2 use table 2 to determine the reaction order for  $\text{na}_2\text{s}_2\text{o}_3$  remember you want to see what happens to the reaction

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