

Practice Final Exam

Pick the best answer for each of the following:

- _____ 1. Which of the following is NOT a true statement concerning what happens in a chemical reaction?
- The ways in which atoms are joined together are changed
 - New atoms are formed as products
 - The starting materials are called reactants
 - The bonds of the reactants are broken and new bonds of the products are formed
 - In a word equation representing a chemical reaction, the reactants are written on the left and the products on the right.

- _____ 2. In the balanced chemical equation for the following reaction, what is the coefficient in front of fluorine?



- a) 1 b) 2 c) 3 d) 5 e) 6

- _____ 3. In order to predict whether or not a redox displacement reaction takes place, we

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- a) measure electron densities
b) orbital energy level diagram

- _____ 4. Predict the products for the following reaction:



- a) potassium zinc + nitrogen
b) potassium nitride + zinc
c) potassium nitrate + zinc
d) potassium nitrate + zinc
e) No Reaction

Lithium	<div style="display: inline-block; width: 10px; height: 40px; background-color: black; position: relative; top: -5px; bottom: -5px;"> ↑ ↓ </div> Activity Series
Potassium	
Calcium	
Magnesium	
Zinc	

- _____ 5. Which of the following accurately represents the correct balanced equation for the following reaction:



- a) $5 \text{ S} + \text{N} \rightarrow \text{N}_5$
b) $5 \text{ S} + \text{N} \rightarrow \text{N}_5\text{S}_5$
c) $3 \text{ S}_8 + \text{N}_2 \rightarrow \text{N}_2\text{S}_6$
d) $6 \text{ N}_2 + \text{S}_8 \rightarrow 2 \text{ N}_2\text{S}_6$
e) $4 \text{ S}_8 + \text{N}_2 \rightarrow 2 \text{ N}_2\text{S}_8$

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Valentine Korah



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