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CHEMISTRY

Page 2

2 hours

INSTRUCTIONS TO CANDIDATES

Section A consists of 10 structured questions. Answer all the questions in this section.

Answers to these questions must be written in the spaces provided.

Section B consists of 6 semi-structured questions. Answer any four questions from this section. Answers to the questions must be written in the answer booklet(s) provided.

In both sections all working must be clearly shown and must be in blue or black ink.

Any work done in pencil will not be marked except drawings.

Mathematical tables and electronic programmable calculators may be used.

Give necessary unit

($g\text{-}1$), ($^{\circ}\text{C}$), (dm^3), (mol), (dm^3), (m^3), (g), (mol), (dm^3), (mol)

1 mole of gas occupies 24 dm³ at room temperature.

1 mole of gas occupies 22.4 dm³ at s.t.p.

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1	2	3	4	5	6	7	8	9	10	11	12	13	14	Total

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UCE 2020

Chemistry

Paper 2

Question 10

Uce Chemistry Paper2 Answers

AW Chickering



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