

Unit 7
Solutions
Chemistry, Grade Level(s): 10
 by Mr. Ma
 12 Days

Stage 1 - Desired Results

Established Goals:

State/Course or Core Standards and school/district-related goals that will be addressed.

- Key Area: 3

Big Ideas (Major Understandings):

(Students will understand that...)

- Mixtures are not compounds.
- Solubility depends on three major factors: temperature, pressure, and the chemical nature of the solute and solvent.
- The boiling and freezing point of a solution is different than that of the pure solvent.

Essential Questions:

What provocative questions will foster inquiry, understanding, and transfer of learning?

- When is something a mixture, and when is it a compound?
- What does the old adage "like dissolves like" mean?
- If CO_2 is normally not soluble in water, then how is it possible to have carbonated drinks?
- Why does salt cause the water to stay in the winter?

Knowledge (important facts):

(Students will know...)

- Matter is classified as a pure substance or as a mixture of substances.
- Mixtures are composed of two or more different substances that can be separated by physical means. When different substances are mixed together, a homogeneous or heterogeneous mixture is formed.
- Differences in properties such as density, particle size, molecular polarity, boiling point and freezing point, and solubility give a physical separation of the components of the mixture.
- A solution is a homogeneous mixture of a solute dissolved in a solvent. The solubility of a solute in a given amount of solvent is dependent on the temperature, the pressure, and nature of the

Skills:

(Students will be able to...)

- describe the process and use of filtration, distillation, and chromatography in the separation of a mixture.
- interpret and construct solubility curves
- use solubility curves to distinguish among saturated, supersaturated, and an unsaturated solutions.

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**Vassilios S. Vassiliadis, E. Antonio del
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