

Chapter – 8

B. Very Short Answer Type Questions:

1. Write the molecular formula of aluminium oxide.

Ans: Al_2O_3

2. The elements present in sodium bicarbonate are sodium, carbon, _____ and oxygen.

Ans: Hydrogen

3. Which one has more mass 1 mole of hydrogen gas or 0.20 mole of nitrogen gas?

Ans: 0.20 mole of nitrogen gas

4. Which one has more number of molecules 1 mole of N_2 or 1 mole of O_2 ?

Ans: Same. Because 1 mole of any substance always contain 6.022×10^{23} of that substance.

5. What is the value of Avogadro's constant?

Ans: 6.022×10^{23}

6. What is the valency of carbon in (a) CCl_4 and (b) CO_2 ?

Ans: (a) CCl_4 – 4, (b) CO_2 – 4

7. If the atomic symbol of Cobalt is Co, then what is the atomic symbol of Copper?

Ans: Cu

8. What is the number of moles in 18g of water?

Ans: 1

9. What is the mass of 1 mole of water?

Ans: 18g

C. Short Answer Questions Type - I:

1. State the Law of constant proportion and name the scientist who proposed this law?

Ans: Law of constant proportion states that,

A chemical compound always contains the same element combined together in the same definite ratio by mass.

This law was proposed by Joseph L. Proust.

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