

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme sollen Sie zur Vertiefung des wissenschaftlichen Verständnisses der Stoffeigenschaften von Metallen herausfordern. Sie sind in der nachfolgenden Tabelle in die Kategorien der Stoffeigenschaften eingeteilt. Die Punkteverteilung ist angegeben.

$$10 \cdot P_1 + 10 \cdot P_2 + 10 \cdot P_3 + 10 \cdot P_4 = 70$$

1. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C.

$$P_1 = 10 \cdot \frac{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}}{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}} = 10 \cdot 0,00017 = 1,7 \text{ Punkte}$$

2. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C.

$$P_2 = 10 \cdot \frac{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}}{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}} = 1,7 \text{ Punkte}$$

3. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C. Die Dichte des Metalls bei 20 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 100 °C?

$$P_3 = 10 \cdot \frac{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}}{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}} = 1,7 \text{ Punkte}$$

4. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C. Die Dichte des Metalls bei 20 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 100 °C? (Hinweis: Die Dichte des Metalls bei 100 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 20 °C?)

$$P_4 = 10 \cdot \frac{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}}{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}} = 1,7 \text{ Punkte}$$

5. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C. Die Dichte des Metalls bei 20 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 100 °C? (Hinweis: Die Dichte des Metalls bei 100 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 20 °C?)

$$P_5 = 10 \cdot \frac{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}}{8,00 \text{ g/cm}^3 \text{ bei } 100^\circ\text{C} - 8,00 \text{ g/cm}^3 \text{ bei } 20^\circ\text{C}} = 1,7 \text{ Punkte}$$

6. Ein bestimmtes Metall (unbekannt) wird bei einer Temperatur von 100 °C erhitzt und bei einer anderen Temperatur abgekühlt. Die gemessene Dichte beträgt 8,00 g/cm³ bei 100 °C. Die Dichte des Metalls bei 20 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 100 °C? (Hinweis: Die Dichte des Metalls bei 100 °C beträgt 8,00 g/cm³. Wie groß ist die Dichte des Metalls bei 20 °C?)

Section 3 Behavior Of Gases Answer Key

Sasha Hu



Section 3 Behavior Of Gases Answer Key:

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Hearings United States. Congress. Senate. Committee on Commerce,

Chance in the World Carl Hoefer, 2019-08-13 Probability has fascinated philosophers, scientists, and mathematicians for hundreds of years. Although the mathematics of probability is for most applications clear and uncontroversial, the interpretation of probability statements continues to be fraught with controversy and confusion. What does it mean to say that the probability of some event X occurring is 31%? In the 20th century, a consensus emerged that there are at least two legitimate kinds of probability, and correspondingly at least two kinds of possible answers to this question of meaning. Subjective probability, also called credence or degree of belief, is a numerical measure of the confidence of some person or some ideal rational agent. Objective probability or chance is a fact about how things are in the world. It is this second type of probability with which Carl Hoefer is concerned in this volume, specifically how we can understand the meaning of statements about objective probability. He aims to settle the question of what objective chances are, once and for all, with an account that can meet the demands of philosophers and scientists alike. For Hoefer, chances are constituted by patterns that can be discerned in the events that happen in our world. These patterns are ideally appropriate guides to what credences limited rational agents such as ourselves should have in situations of imperfect knowledge. By showing this, Hoefer bridges the gap between subjective probability and chance. In a field where few scholars have given adequate treatment to interpreting statements of chance, Hoefer develops a philosophically rich theory which draws on the disciplines of metaphysics, ontology, and philosophy of science.

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