

Section 8.4 Strength of Acids and Bases

- Weak Acids
- Concentration-the amount of solute dissolved in a given amount of solution
- Strength refers to: the solute's tendency to form ions in water
- You cannot assume that a strong acid has a low pH, because its concentration also affects pH.
- Weak Base: Ammonia

Section 8 4 Strength Of Acids And Bases

JR Anderson



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