

The Periodic Table and Chemical Properties

Textbook pages 52–63

Before You Read

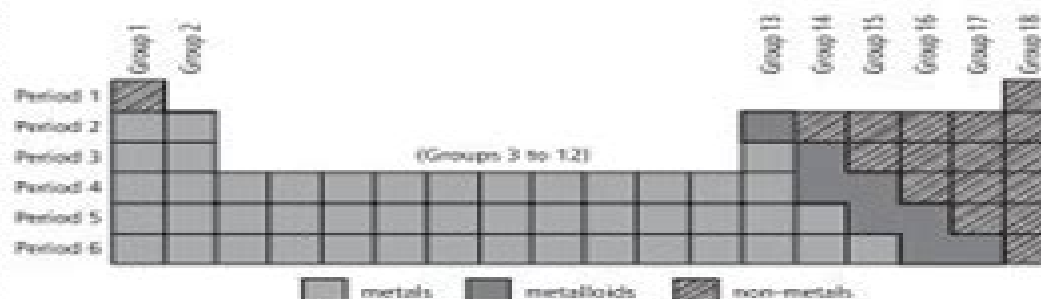
Imagine you had to classify all of the different elements. How would you organize them? How might properties help you decide? Write your thoughts below.



Mark the Text

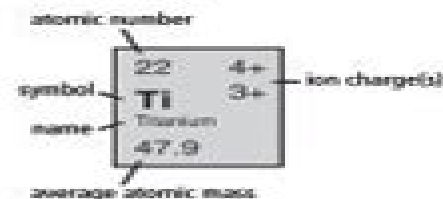
Identify Details

As you skim the section, use one colour to highlight the text or labels that talk about the periodic table. Use another colour to highlight facts about atomic number and other facts about each element.



What is the periodic table?

The **periodic table** places each element on a chart based on its chemical and physical properties. Each box of the periodic table lists information about the atoms that make up an element. You can see:



- ◆ the name of the element
- ◆ the chemical symbol of the element
- ◆ the **atomic number** of the element—the number of protons in the nucleus of each one of its atoms
- ◆ the **average atomic mass** of the element—the weighted average of the masses of the atoms of an element
- ◆ the **ion charge** (or charges) of the element—the electric charge of its atoms when they gain or lose electrons. If the atoms can gain or lose electrons in more than one way, they will have a **multiple ion charge**. ➔



Reading Check

1. What five facts about an element can you find in the periodic table?

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Walter Henry Lange



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