

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 - (a) how many atoms of Fe are in one formula unit
 - (b) how many sulfate ions can be attached to the iron atom
 - (c) the charge on each Fe ion
 - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to _____.
 - (a) change the number of moles represented by the formula
 - (b) change the charges on the other ions in the compound
 - (c) change the formula so that it no longer represents the compound it previously represented
 - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.

<u>4 elements</u>	a. How many elements make up this compound?
<u>6 oxygen atoms</u>	b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
<u>21 atoms</u>	c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
<u>4.2×10^{24} atoms</u>	d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
- 4.** How many atoms are present in each of these formula units?

<u>11 atoms</u>	a. $Ca(HCO_3)_2$
<u>45 atoms</u>	b. $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	c. $Fe(ClO_2)_3$
<u>9 atoms</u>	d. $Fe(ClO_3)_3$
- 5.**

<u>N_2O_5</u>	a. What is the formula for the compound dinitrogen pentoxide?
<u>carbon(IV) sulfide</u>	b. What is the Stock name for the covalent compound CS_2 ?
<u>H_2SO_3</u>	c. What is the formula for sulfurous acid?
<u>phosphoric acid</u>	d. What is the name for the acid H_3PO_4 ?

Section 6 Chemical Names And Formulas Key

M Mosston



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