

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Section 6 Review Chemical Bonding Answer Key

Arshad Iqbal



Section 6 Review Chemical Bonding Answer Key:

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Sulfide Mineralogy and Geochemistry David J. Vaughan, 2018-12-17. Volume 61 of Reviews in Mineralogy and Geochemistry presents an up-to-date review of sulfide mineralogy and geochemistry. The crystal structures, electrical and magnetic properties, spectroscopic studies, chemical bonding, thermochemistry, phase relations, solution chemistry, surface structure, and chemistry of hydrothermal precipitation processes of sulfur isotopes, geochemistry, and geobiology of metal sulfides are reviewed. Where appropriate for comparison, there is a brief discussion of the selenide or telluride analogs of the metal sulfides. When discussing crystal structures and structural relationships, the sulfosalt minerals as well as the sulfides are considered in some detail. **Anatomy & Physiology (includes A&P Online course) E-Book** Kevin T. Patton, 2018-01-31. Anatomy Physiology includes A&P Online course E-Book. **McGraw-Hill Education SAT Subject Test Chemistry, Fifth Edition** Thomas A. Evangelist, 2018-11-23. We Will Help You Get Your Best Score. With more than 125 years of experience in education, McGraw-Hill Education is the name you trust to deliver results. This MHE guide is the most comprehensive and relevant SAT Subject Test prep tool on the market. This edition provides 5 full-length practice tests with thorough answer explanations. A comprehensive review of all Chemistry concepts essential to success on the SAT Subject Test. An extensive overview of the format of the test based on the most recent SAT Chemistry exams. Unique test-taking strategies and tips recommended by teachers to help you raise your score. A customizable study plan to help you maximize the time you have to prepare. **TOP 40 LIST** The book includes a description of the 40 topics that are most crucial to know before you take the Subject Test in Chemistry. **TEST TAKING STRATEGIES** Learn unique tips developed by teachers to help you avoid the test maker's traps. Nuclear Magnetic Resonance G. A. Webb, 2007-10-31. As a spectroscopic method, Nuclear Magnetic Resonance (NMR) has seen spectacular growth over the past two decades, both as a technique and in its applications. Today the applications of NMR span a wide range of scientific disciplines from physics to biology to medicine. Each volume of Nuclear Magnetic Resonance comprises a combination of annual and biennial reports which together provide comprehensive coverage of the literature on this topic. This Specialist Periodical Report reflects the growing volume of published work involving NMR techniques and applications in particular, NMR of natural macromolecules, which is covered in two reports: NMR of Proteins and Acids and NMR of Carbohydrates, Lipids, and Membranes. For those wanting to become rapidly acquainted with specific areas of NMR, this title provides unrivalled scope of coverage. Seasoned practitioners of NMR will find this an invaluable source of current methods and applications. Specialist Periodical Reports provide systematic and detailed review coverage in major areas of chemical research. Compiled by teams of leading authorities in the relevant subject areas, the series creates a

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obtained my undergraduate and postgraduate degrees in the 1970s It was during this period that I began my continuing love
affair with this subject and the fact that it happened while I was a student in an organic laboratory is beside the point I was
always enchanted by the more physical aspects of inorganic chemistry while being captivated from an early stage by the
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