

Describing Matter by using **extensive** and **intensive properties**



Extensive Properties are properties that depend on the amount of matter in a sample. They include **mass** and **volume**. The **mass** of an object is a measure of the amount of matter the object contains. The **volume** of an object is a measure of the space occupied by the object.

Intensive properties depend on the type of matter in a sample, not the amount of matter. The hardness of a boiling ball is an example of an intensive property.

Section Describing Matter Prentice Hall

Michael Rajnik



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Prentice Hall Science Explorer Michael J. Padilla, 2002 *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall, Michael J. Padilla, 2004-10 *Addison-Wesley Introduction to Physical Science* Michael B. Leyden, 1988 *Chapter 3: Matter, Energy and the Universe* Michael Pidwirny, 2023-06-28

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Wiley FRM Exam Review Study Guide 2016 Part I Volume 2 Wiley,2016-01-19 **Prentice Hall Physical Science Concepts in Action Program Planner**

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