

Section 7.4 - Determining Chemical Formulas

Calculations of Molecular Formulas

An empirical formula may or may not be a correct molecular formula.

Book example , diborane¹'s empirical formula is BH_3 , any multiple of that equals the same ratio - $\text{B}_2\text{H}_6, \text{B}_3\text{H}_9, \text{B}_4\text{H}_{12}$ etc

It is a colorless gas at room temperature with a repulsively sweet odor.
Diborane mixes well with air, easily forming explosive mixtures.
Diborane will ignite spontaneously in moist air at room temperature.

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