Name: Kegan Graves Date:

## Student Exploration: Limiting Reactants

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

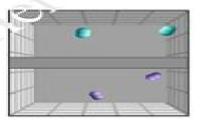
Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

- How many total hot dogs can you make?
- 3. Which ingredient will you have leftovers of? hotdog:

#### Gizmo Warm-up

Just as ingredients can be put together to make a new rod, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reaction will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting reactants*. Sizmo<sup>198</sup>, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $H_2 + C_2$  becomes  $E_2O$  is salected. The small "2" in  $H_2$ ,  $O_2$ , and  $H_2O$  is a subscript. Subscripts represent the number of atoms in a **molecule**.

- Use the sliders to set the number of O<sub>2</sub> molecules and H<sub>2</sub> molecules to two.
  - A. How many nydrogen molecules (H<sub>2</sub>) are there? \_\_2\_\_\_ Hydrogen
  - B. Fow niany oxygen molecules (O<sub>2</sub>) are there? \_2\_\_\_\_ Oxygen
- How many H<sub>2</sub>O molecules do you think will form when these four molecules react?
- Click Play ( ). How many H<sub>2</sub>O molecules actually formed? \_\_\_\_\_3\_



# **Student Exploration Limiting Reactants Explore Learning Teacher Key**

**Ying-Ying Zheng** 

**Student Exploration Limiting Reactants Explore Learning Teacher Key:** 

Unveiling the Magic of Words: A Review of "Student Exploration Limiting Reactants Explore Learning Teacher Key"

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