

Name: Date:

Student Exploration: Limiting Reactants

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?

2. Which ingredient limited the number of hot dogs you could make?

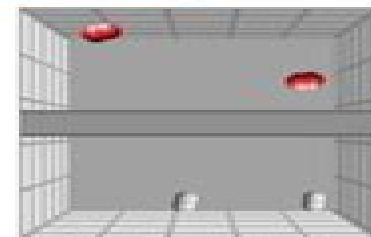
3. Which ingredient will you have leftovers of?

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure $\text{H}_2 + \text{O}_2$ becomes H_2O is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.



1. Use the sliders to set the number of O_2 molecules and H_2 molecules to two.

A. How many hydrogen molecules (H_2) are there?

Hydrogen atoms?

B. How many oxygen molecules (O_2) are there?

Oxygen atoms?

2. How many H_2O molecules do you think will form when these four molecules react?

3. Click **Play** (). How many H_2O molecules actually formed?

Student Exploration Limiting Reactants Key

Lifelique



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ChemDiscovery Student Guide Olga I. Agapova, 2002 **Prentice Hall Chemistry**, 2000 *Computers in Education*, 1975 **Limiting Reactant and Percent** Lifeliqe, 2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

Adopting the Song of Expression: An Emotional Symphony within **Student Exploration Limiting Reactants Key**

In a global consumed by screens and the ceaseless chatter of instantaneous conversation, the melodic splendor and emotional symphony created by the prepared word often disappear in to the back ground, eclipsed by the persistent noise and interruptions that permeate our lives. But, nestled within the pages of **Student Exploration Limiting Reactants Key** an enchanting fictional value filled with natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an outstanding musician of language, this interesting masterpiece conducts viewers on a mental trip, skillfully unraveling the concealed tunes and profound influence resonating within each cautiously crafted phrase. Within the depths of the emotional examination, we will discover the book is key harmonies, analyze their enthralling publishing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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