

Name: Brittany French

Date: October 3, 2012

## Student Exploration: Electron Configuration

**Vocabulary:** atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

### Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo™*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? **3**

Atomnummer  
Elementsymbol  
Atomgewicht  
chemisches Element

→

→

3
Li
6.94

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100																							
H	He	Li	Be	B	C	N	O	F	Ne	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120

2. A neutral atom has the same number of electrons and protons.

How many electrons are in a neutral lithium atom? **3**

3. Select the **ELECTRON CONFIGURATION** tab. Click twice in the **1s** box at upper left and once in the **2s** box. Observe the atom model on the right.

A. What do you see? **I see 2 electrons in the first shell and 1 in the second orbiting the nucleus.**

B. Click **Check**. Is this **electron configuration** correct? **Yes, the electron configuration is correct.**

**Activity A:**  
**Small atoms**

Get the Gizmo ready:

- On the PERIODIC TABLE tab, select **H** (hydrogen).
- Select the **ELECTRON CONFIGURATION** tab.
- Click **Reset**.

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100																							
H	He	Li	Be	B	C	N	O	F	Ne	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120

**Introduction:** Electrons are arranged in **orbitals**, **subshells**, and **shells**. These levels of organization are shown by the boxes of the Gizmo. Each box represents an orbital. The subshells are labeled with letters (*s*, *p*, *d*, and *f*) and the shells are labeled with numbers.

**Question:** How are electrons arranged in elements with atomic numbers 1 through 10?

1. **Infer:** Based on its atomic number, how many electrons does a hydrogen atom have? **1**

# Student Exploration Electron Configuration Answer Key Gizmo

**E Durkheim**



## **Student Exploration Electron Configuration Answer Key Gizmo:**

## **Student Exploration Electron Configuration Answer Key Gizmo** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Student Exploration Electron Configuration Answer Key Gizmo**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents Student Exploration Electron Configuration Answer Key Gizmo**

1. Understanding the eBook Student Exploration Electron Configuration Answer Key Gizmo
  - The Rise of Digital Reading Student Exploration Electron Configuration Answer Key Gizmo
  - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Electron Configuration Answer Key Gizmo
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Student Exploration Electron Configuration Answer Key Gizmo
  - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Electron Configuration Answer Key Gizmo
  - Personalized Recommendations
  - Student Exploration Electron Configuration Answer Key Gizmo User Reviews and Ratings
  - Student Exploration Electron Configuration Answer Key Gizmo and Bestseller Lists

5. Accessing Student Exploration Electron Configuration Answer Key Gizmo Free and Paid eBooks
  - Student Exploration Electron Configuration Answer Key Gizmo Public Domain eBooks
  - Student Exploration Electron Configuration Answer Key Gizmo eBook Subscription Services
  - Student Exploration Electron Configuration Answer Key Gizmo Budget-Friendly Options
6. Navigating Student Exploration Electron Configuration Answer Key Gizmo eBook Formats
  - ePub, PDF, MOBI, and More
  - Student Exploration Electron Configuration Answer Key Gizmo Compatibility with Devices
  - Student Exploration Electron Configuration Answer Key Gizmo Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Student Exploration Electron Configuration Answer Key Gizmo
  - Highlighting and Note-Taking Student Exploration Electron Configuration Answer Key Gizmo
  - Interactive Elements Student Exploration Electron Configuration Answer Key Gizmo
8. Staying Engaged with Student Exploration Electron Configuration Answer Key Gizmo
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Student Exploration Electron Configuration Answer Key Gizmo
9. Balancing eBooks and Physical Books Student Exploration Electron Configuration Answer Key Gizmo
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Student Exploration Electron Configuration Answer Key Gizmo
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Electron Configuration Answer Key Gizmo
  - Setting Reading Goals Student Exploration Electron Configuration Answer Key Gizmo
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Electron Configuration Answer Key Gizmo
  - Fact-Checking eBook Content of Student Exploration Electron Configuration Answer Key Gizmo
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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