Name: Brittany French

Date: October 3, 2012

# Student Exploration: Electron Configuration

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

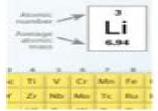
### Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the Electron Configuration Gizmo™.

To begin, check that Lithium is selected on the PERIODIC TABLE tab.

The atomic number is equal to the number of protons in an atom.

How many protons are in a lithium atom? 3



A neutral atom has the same number of electrons and protons.

How many electrons are in a neutral lithium atom? 3

- Select the ELECTRON CONFIGURATION tab. Click twice in the 1s box at upper left and once in the 2s box. Observe the atom model on the right.
  - A. What do you see? I see 2 electrons in the first shell and 1 in the second orbiting the nucleus.
  - B. Click Check. Is this electron configuration correct? Yes, the electron configuration is correct.

	Get the Gizmo ready:	
Activity A: Small atoms	On the PERIODIC TABLE tab, select H (hydrogen).     Select the ELECTRON CONFIGURATION tab.     Click Reset.	H 2

**Introduction:** Electrons are arranged in **orbitals**, **subshells**, and **shells**. These levels of organization are shown by the boxes of the Gizmo. Each box represents an orbital. The subshells are labeled with letters  $(s, \rho, d, and f)$  and the shells are labeled with numbers.

Question: How are electrons arranged in elements with atomic numbers 1 through 10?

Infer: Based on its atomic number, how many electrons does a hydrogen atom have?



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