

Chapter 4 – Alkanes

4.1 Nomenclature of Alkanes

RULES FOR NAMING ALKANES

- 1 Identify the longest continuous carbon chain (a.k.a the parent chain).
-If there are multiple ways to come up with the longest chain, then choose the one with the most substituents as the parent chain.
- 2 Number the carbons in the parent chain starting on the end closest to the first substituent. If there is a tie, then move on to the second substituent and so on until there is not a tie.
-If the substituents are ALL located at the same chain locations in the parent chain when numbered in both directions, then compare the 1st substituent encountered in both directions and rank by alphabetical order. If there is a tie, then move on to the second substituent and so on until there is not a tie.
- 3 Identify the substituents attached to the parent chain. Carbon substituents end with the suffix -yl (methyl, ethyl, propyl, etc.).
-For multiple identical substituents use the prefix di-, tri-, tetra- etc. Use the chain number (and a hyphen) as the locator before each substituent. For multiple identical substituents list all chain locators separated by commas.
- 4 List substituents in alphabetical order. In determining alphabetical order ignore numerical prefixes and hyphenated prefixes (*tert*- and *sec*-), but not *iso* and *cyclo*.
-Use the chain number (and a hyphen) as the locator before each substituent.
-For multiple identical substituents use the prefixes di-, tri-, tetra- etc.
- For multiple identical substituents list all chain locators separated by commas.
- 5 The name ends with the name of the parent chain. There should NOT be a space between the substituents and the name of the parent chain.

Prefixes

1	meth
2	eth
3	prop
4	but
5	pent
6	hex
7	hept
8	oct
9	non
10	dec
11	undec
12	dodec

Identify and number the parent chain in each of the following. Then provide the IUPAC name.



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Table of Contents Study Guide Alkanes Answers Keys

1. Understanding the eBook Study Guide Alkanes Answers Keys
 - The Rise of Digital Reading Study Guide Alkanes Answers Keys
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Alkanes Answers Keys
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Alkanes Answers Keys
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Alkanes Answers Keys

- Personalized Recommendations
- Study Guide Alkanes Answers Keys User Reviews and Ratings
- Study Guide Alkanes Answers Keys and Bestseller Lists
- 5. Accessing Study Guide Alkanes Answers Keys Free and Paid eBooks
 - Study Guide Alkanes Answers Keys Public Domain eBooks
 - Study Guide Alkanes Answers Keys eBook Subscription Services
 - Study Guide Alkanes Answers Keys Budget-Friendly Options
- 6. Navigating Study Guide Alkanes Answers Keys eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide Alkanes Answers Keys Compatibility with Devices
 - Study Guide Alkanes Answers Keys Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide Alkanes Answers Keys
 - Highlighting and Note-Taking Study Guide Alkanes Answers Keys
 - Interactive Elements Study Guide Alkanes Answers Keys
- 8. Staying Engaged with Study Guide Alkanes Answers Keys
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Alkanes Answers Keys
- 9. Balancing eBooks and Physical Books Study Guide Alkanes Answers Keys
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Alkanes Answers Keys
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide Alkanes Answers Keys
 - Setting Reading Goals Study Guide Alkanes Answers Keys
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide Alkanes Answers Keys

- Fact-Checking eBook Content of Study Guide Alkanes Answers Keys
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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