

Honors Biology
II semester final examination
Study Guide

Chapters for final examinations:35,36,37,38,39,40.

Read the following concepts for short and essay questions:

Chapter 35:

1. Human organ systems.
2. Types of tissues
3. Types of epithelial tissues.
4. Maintaining of Homeostasis in the body.
5. Structure of Neuron. (Label the Diagram)
6. How nerve impulse is generated.
7. Define membrane, threshold, action potential.
8. The synapse.
9. Central nervous system and its parts and functions.
10. Difference between somatic and autonomic nervous system.
11. Structure of eye and ear for labeling.

Chapter 36:

1. Label Axial and Appendicular skeleton.
2. Describe the structure of bone.
3. Name the types of joints and give examples.
4. Name the types of muscle tissues, explain their structure and function.
5. What is integumentary system.

Chapter 37:

1. Describe the structure of heart, and label the parts.
2. What is double circulation.
3. What is atherosclerosis.
4. Name the blood cells and their functions.
5. What is lymphatic system, how it is useful for the body.

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Zimmerman, Sophie E. Merritt, 1963 **Directory of Distance Learning Opportunities** Modoc Press, Inc., 2003-02-28 This
book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via
such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154
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senior projects The book begins with a math and physics review to ensure all students start on the same level and then
discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It
then provides material for an optional second semester course that shows students how to apply their enhanced
mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the
text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom
tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key
concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn
mathematical derivations Mini biographies of famous scientists make the presentation more interesting from a people point
of view Stating the basic concepts of quantum chemistry in terms of analogies provides a pedagogically useful technique
Covering key topics such as the critical point of a van der Waals gas the Michaelis Menten equation and the entropy of

mixing this classroom tested text highlights applications across the range of chemistry forensic science pre medical science and chemical engineering In a presentation of fundamental topics held together by clearly established mathematical models the book supplies a quantitative discussion of the merged science of physical chemistry *Announcements* Westmar College (Le Mars, Iowa),1929 *Resources in Education* ,1995 *Announcements for the Year ...* Purdue University,1912 *The Annual Catalogue of Purdue University, Lafayette, Indiana ... with Announcements for ...* Purdue University,1917 The Earlham College Bulletin Earlham College,1929 **Compilation from the Annual Reports of the Superintendent of Public Instruction of the State of Michigan, for the Years 1845 and '46, with Important Additions, Embracing the Report for the Year 1847** Michigan. Department of Public Instruction,1892 *Announcements* University of Nebraska (Lincoln campus). Teachers College,1924 *Circular* ,1922

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