

Chemical Names and Formulas

❖ Key Vocabulary

Monoatomic Ion – Ions formed from a single atom

Binary compound – Compounds composed of two elements

Nomenclature – the “naming system” of binary ionic compounds

Oxyanion – polyatomic ions that contain oxygen

Salt – an ionic compound composed of a cation and the anion from an acid

Chemical Formulas

Subscripts after an elemental symbol represent the number of atoms of the element in the compound

Molecular formulas represent how many atoms of each element are in a compound.
(C_2H_6 , H_2O , $C_6H_{12}O_6$)

Ionic formulas show the simplest ratio, cations to anions, of atoms in the lattice

Parenthesis are used around a polyatomic ion in a chemical formula

Examples $Al_2(SO_4)_3$, $Cu(PO_4)_2$, $Sr(NO_3)_2$

Monoatomic Ions

– Ions formed from a single atom

Group 1 elements may lose one electron = 1+

Group 2 elements may lose two electrons = 2+

Group 13 elements may lose three electrons = 3+

Group 17 elements may gain one electron = 1-

Group 16 elements may gain two electrons = 2-

Group 15 elements may gain three electrons = 3-

Transition metals have various possibilities (no pattern)

Naming monatomic Ions

Monatomic cations are simply called by the element's name

*the stock system is used when there is more than one possibility for an element (more on this later)

Monatomic anions end in IDE such as Fluorine – Fluoride, Oxygen – Oxide
(reference table 1 on page 221)

Binary Ionic Compounds

1. Write the symbols for the ions with the cation first (Al^{3+} O^{2-})

2. Cross over the charges writing the charge of one ion as the subscript for the other ion (Al_2^{3+} O_3^{2-})

3. Multiply subscript and superscript for each ion then find the lowest whole number ratio (Al_2O_3)

Name it!!! Aluminum Oxide

Examples: Zn^{2+} I^- Zn^{2+} S^{2-}

(More on p 223 practice problems)

Study Guide Chemical Names And Formulas

James J. Mizner



Study Guide Chemical Names And Formulas:

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011 **Student Study Guide/Solutions**

Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15

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from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **Study Guide for Chang's Chemistry** Kenneth W. Watkins, 1988

Scientific Information Notes , 1961-10 *Study Questions in Elementary Organic Chemistry* Alexander Lowy, 1926

Student Guide For Living Chemistry Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student's understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable **General, Organic, and Biochemistry Study Guide** Marcia L. Gillette, Ira Blei, Wendy Gloffke, George Odian, 2006-01-27 This study guide provides reader friendly reinforcement of the concepts covered in the textbook Features include Chapter outlines Are you able to Worked text problems Fill ins Test yourself Concept maps Can also be used for Blei and Odian's Organic and Biochemistry **Study Guide [to] Chemistry, a Systematic Approach [by] Sisler, Dresdner, and Mooney** William H. Myers, William T. Mooney, 1980 **Barron's Chemistry Practice Plus: 400+ Online Questions and Quick Study Review** Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2022-07-05 Need quick review and practice to help you excel in Chemistry Barron's Chemistry Practice Plus features more than 400 online practice questions and a concise review guide that covers the basics of Chemistry Inside you'll find Concise review on the basics of Chemistry an excellent resource for students who want a quick review of the most important topics Access to 400 online questions arranged by topic for customized practice Online practice includes answer explanations with expert advice for all questions plus scoring to track your progress This essential guide is the perfect practice supplement for students and teachers General Chemistry, Study Guide James E. Brady, 1990-03-01

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Directions and Study Questions in Inorganic Chemistry Alexander Silverman, 1919 Drug & Chemical Markets , 1925

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **Student Notebook and Study Guide to Accompany The Human Body** Bruce Wingerd, Patty Bostwick Taylor, 2013-02-01 This Student Notebook and Study Guide the ideal companion to Bruce Wingerd's The Human

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