

## Chemical Bonding

### SECTION 3

**SHORT ANSWER** Answer the following questions in the space provided.

- A** The notation for sodium chloride, NaCl, stands for one  
 (a) formula unit. (c) crystal.  
 (b) molecule. (d) atom.
- D** In a crystal of an ionic compound, each cation is surrounded by a number of  
 (a) molecules. (c) dipoles.  
 (b) positive ions. (d) negative ions.
- B** Compared with the neutral atoms involved in the formation of an ionic compound, the crystal lattice that results is  
 (a) higher in potential energy. (c) equal in potential energy.  
 (b) lower in potential energy. (d) unstable.
- B** The lattice energy of compound A is greater in magnitude than that of compound B. What can be concluded from this fact?  
 (a) Compound A is not an ionic compound.  
 (b) It will be more difficult to break the bonds in compound A than those in compound B.  
 (c) Compound B has larger crystals than compound A.  
 (d) Compound A has larger crystals than compound B.
- B** The forces of attraction between molecules in a molecular compound are generally  
 (a) stronger than the attractive forces among formula units in ionic bonding.  
 (b) weaker than the attractive forces among formula units in ionic bonding.  
 (c) approximately equal to the attractive forces among formula units in ionic bonding.  
 (d) equal to zero.
- Describe the force that holds two ions together in an ionic bond.

**Very strong overall attraction between positive and negative charges**

- What type of energy best represents the strength of an ionic bond?

**Lattice energy**

- What types of bonds are present in an ionic compound that contains a polyatomic ion?

# Section 3 3 4 Chemistry Of Cells Worksheet

**Barbara Thomas, Laura Matthews**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the authors' names.

### Section 3 3 4 Chemistry Of Cells Worksheet:

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- Chapter 1: Introduction to Section 3 3 4 Chemistry Of Cells Worksheet
- Chapter 2: Essential Elements of Section 3 3 4 Chemistry Of Cells Worksheet
- Chapter 3: Section 3 3 4 Chemistry Of Cells Worksheet in Everyday Life
- Chapter 4: Section 3 3 4 Chemistry Of Cells Worksheet in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Section 3 3 4 Chemistry Of Cells Worksheet. This chapter will explore what Section 3 3 4 Chemistry Of Cells Worksheet is, why Section 3 3 4 Chemistry Of Cells Worksheet is vital, and how to effectively learn about Section 3 3 4 Chemistry Of Cells Worksheet.
3. In chapter 2, this book will delve into the foundational concepts of Section 3 3 4 Chemistry Of Cells Worksheet. This chapter will elucidate the essential principles that need to be understood to grasp Section 3 3 4 Chemistry Of Cells Worksheet in its entirety.
4. In chapter 3, this book will examine the practical applications of Section 3 3 4 Chemistry Of Cells Worksheet in daily life. This chapter will showcase real-world examples of how Section 3 3 4 Chemistry Of Cells Worksheet can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Section 3 3 4 Chemistry Of Cells Worksheet in specific contexts. This chapter will explore how Section 3 3 4 Chemistry Of Cells Worksheet is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Section 3 3 4 Chemistry Of Cells Worksheet. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Section 3 3 4 Chemistry Of Cells Worksheet.

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