

Forming Chemical Bonds

Section 8.1

- I. Chemical Bonds
 - a. The force that holds two atoms together is called a chemical bond.
 - b. Chemical bonds may form by the attraction between a positive nucleus and negative electrons or the attraction between a positive ion and a negative ion.
 - c. Valence electrons are involved in the formation of chemical bonds between two atoms.
 - d. How do thousands of compounds form from the relatively few elements known to exist?
 - i. The answer lies in the electron structure of the atoms of the elements involved and the nature of the attractive forces between these atoms.
 - ii. Recall that electron-dot structure is a type of diagram used to keep track of valence electrons and is especially useful when illustrating the formation of chemical bonds.
 - iii. Recall that ionization energy refers to how easily an atom loses an electron. The term electron affinity indicates how much attraction an atom has for electrons.
 - iv. The difference in reactivity is directly related to the valence electrons.
- II. Formation of Positive Ions
 - a. Compare sodium and neon electron configuration.
 - b. Compare sodium ion and neutral sodium.
 - c. It is possible for atoms to achieve a *pseudo-noble gas configuration*.
 - i. Zinc electron configuration?
 1. What electrons would it lose?
 - d. A positively charged ion is called a **cation**.
- III. Formation of Negative Ions
 - a. Recall nonmetals have a great attraction for electrons and form a stable outer configuration by gaining electrons.
 - b. Chlorine electron configuration?
 - i. What happens?
 - ii. What noble gas does it become like?
 - c. A negatively charged ion is called an **anion**.

The Formation and Nature of Ionic Bonds

Section 8.2

- I. Formation of an Ionic Bond
 - a. The electrostatic force that holds oppositely charged particles together in an ionic compound is referred to as an **ionic bond**.
- II. Compounds that contain ionic bonds are ionic compounds.
 - a. If ionic bonds occur between metals and nonmetal oxygen, oxides form.
 - b. Most other ionic compounds are called salts.
 - c. Binary ionic compounds contain a metallic cation and nonmetallic anion.
- III. What happens in the formation of a compound?
 - a. Explain the formation of the ionic compound composed of each pair of elements.
 - i. sodium and nitrogen

Section 8 1 Forming Chemical Bonds Study Guide

Neil E. Schore



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Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer *Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter*, 1993 Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples *Organic Chemistry, 4e Student Solution Manual and Study Guide* David R. Klein, 2021-01-07 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Solution Manual and Study Guide 4th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Student Study Guide to Accompany Chemistry Kenneth J. Hughes, Paul B. Kelter, 2002-05 Reviews essential points from the text presents problems and hints to assist the student in solving them Solutions to selected problems in the end of chapter problem sets are included By Kenneth Hughes of the University of Wisconsin Madison **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate

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